

Astrochemistry

Lecture 4



*Diffuse and translucent clouds,
PDRs*

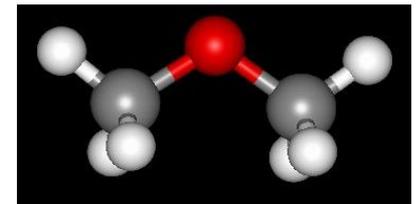
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Leiden Observatory

May-June 2022



Van Dishoeck 1998, in MA II
Snow & McCall 2006, Gerin et al. 2016 ARA&A
Tielens Chap. 9



Summary lectures 1+2

- Large variety of molecules observed in ISM
- Basic processes for formation and destruction identified
- Networks built for explaining abundances (lectures 3 and 4)
- Results depend on thousands of input rates
 - Many not known under astrophysical conditions (but for many accurate rates not needed)
 - Key: identify those reactions which are important to study well
- Many experiments and theory on basic processes over last 30 years
 - New chemical physics insight
 - Significant progress in neutral-neutral reactions, surface reactions
- Some processes now well understood, others take decades of hard work to make just a little progress
 - Funding of lab astrophysics groups issue
 - Note: photoionization codes got atomic input data thanks to a century of efforts in atomic and plasma physics

Summary Lecture 3

- Chemistry starts at era of recombination $z \sim 1000$
 - Only H, D He, Li, no dust
 - First molecules He_2^+ and HeH^+
- Small fraction of H_2 forms through H^+ and H^- routes
 - HD from reactions of D^+ with H_2
- H_2 and HD important coolants of primordial clouds and thus their fragmentation
 - Determines mass of the first stars
- Observations of molecules now up to $z \sim 7$

Outline

A. Diffuse and translucent clouds

- Introduction
- Physical structure
- Gas-phase chemistry networks
- Steady-state models: $\text{H} \rightarrow \text{H}_2$, $\text{C}^+ \rightarrow \text{CO}$

B. Photon-dominated regions (PDRs)

- Introduction
- Chemical structure
- Examples: from Orion to galaxies at low metallicity to protoplanetary disks

A. Diffuse and translucent clouds

4.1 Introduction

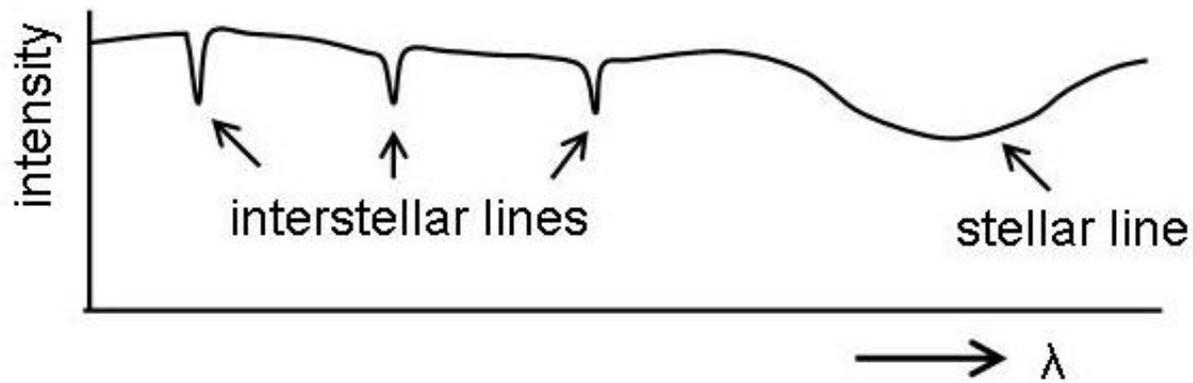
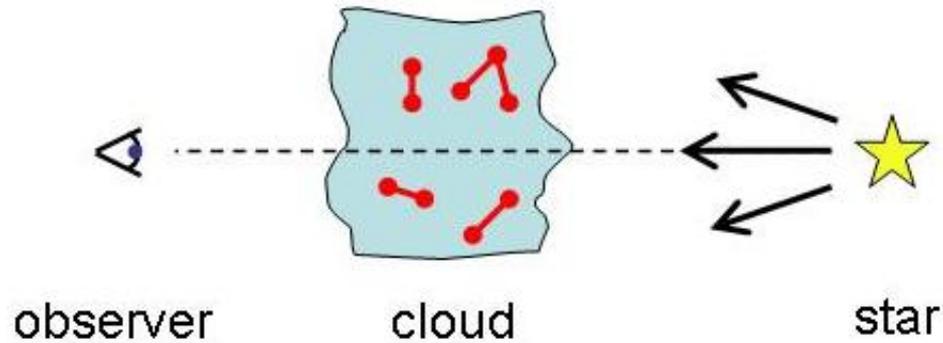
- Diffuse clouds with $A_V \approx 1$ mag are the best testing ground for basic interstellar chemistry because

Mostly

- ~~Only~~ simplest diatomic molecules found
- Gas-phase elemental abundances directly measured
- H and H₂ directly measured
- UV penetrates \Rightarrow short timescales \Rightarrow chemical equilibrium
- No embedded / hidden energy sources
- Models first considered by Kramers & ter Haar (1946) and Bates & Spitzer (1951). Subsequent models include ion-molecule network developed by Herbst & Klemperer (1973)

Recall relation A_V and total column density N_H

Observations of diffuse clouds



Questions

- To what extent can ion-molecule gas-phase chemistry reproduce observed abundances?
- What is role of shocks and turbulence?
- What is role of grains in formation molecules other than H₂?
- What is relation diffuse – dense clouds?
⇒ study also *translucent** clouds with
 $A_V=1-5$ mag

* Term coined by van Dishoeck & Black (1989)

Observations diffuse clouds

VIS + UV absorption lines

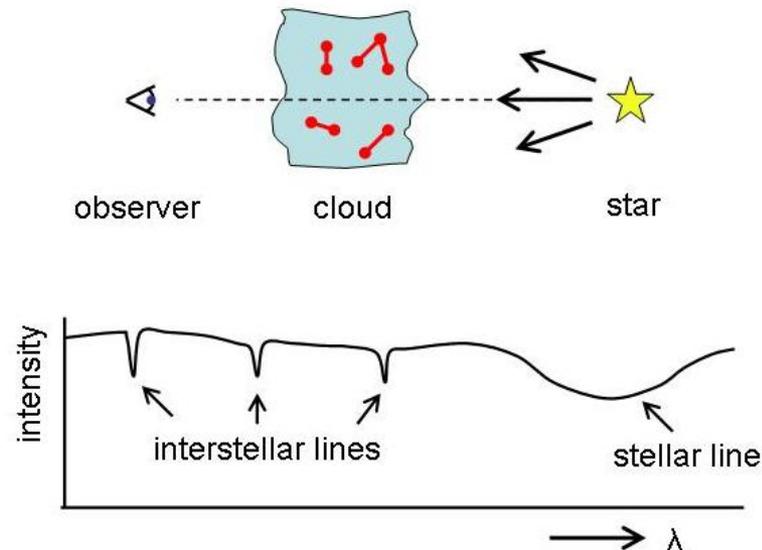
- Many atomic lines \Rightarrow depletion of elements C, O, N, ..
 - Chemistry sensitive to C/O ratio
- Few molecular lines detected
 - H₂, HD
 - CH, CH⁺, C₂, C₃
 - OH, OH⁺, CO
 - NH, CN, HCl
- Not detected at optical wavelengths:
 - H₂O, H₂O⁺, C₄, ...
 - MgH, NaH, SH⁺, ...

Millimeter/far-IR absorption and emission lines

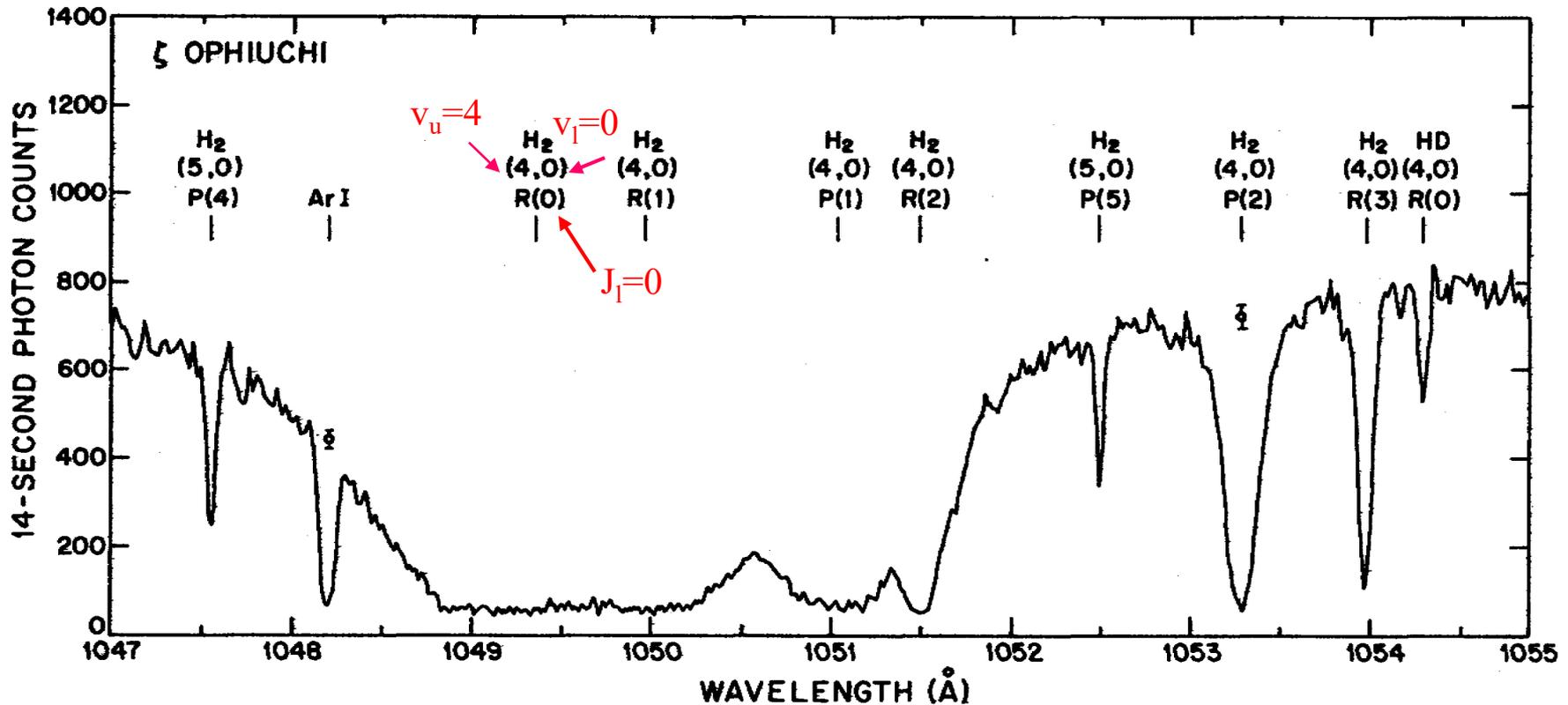
- Weak emission, many absorption lines

Observations of diffuse clouds

- Observed primarily by absorption lines at visible (since 1900's) and UV wavelengths (since 1970's: *Copernicus, FUSE, HST*)
- Classical example: line-of-sight toward ζ Oph
- Spectra show sharp interstellar lines super-imposed on broad stellar lines



Interstellar H₂ lines towards ζ Ophiuchi



Jenkins et al., Copernicus satellite

$\Delta J = J_u - J_l = +1$: R-branch

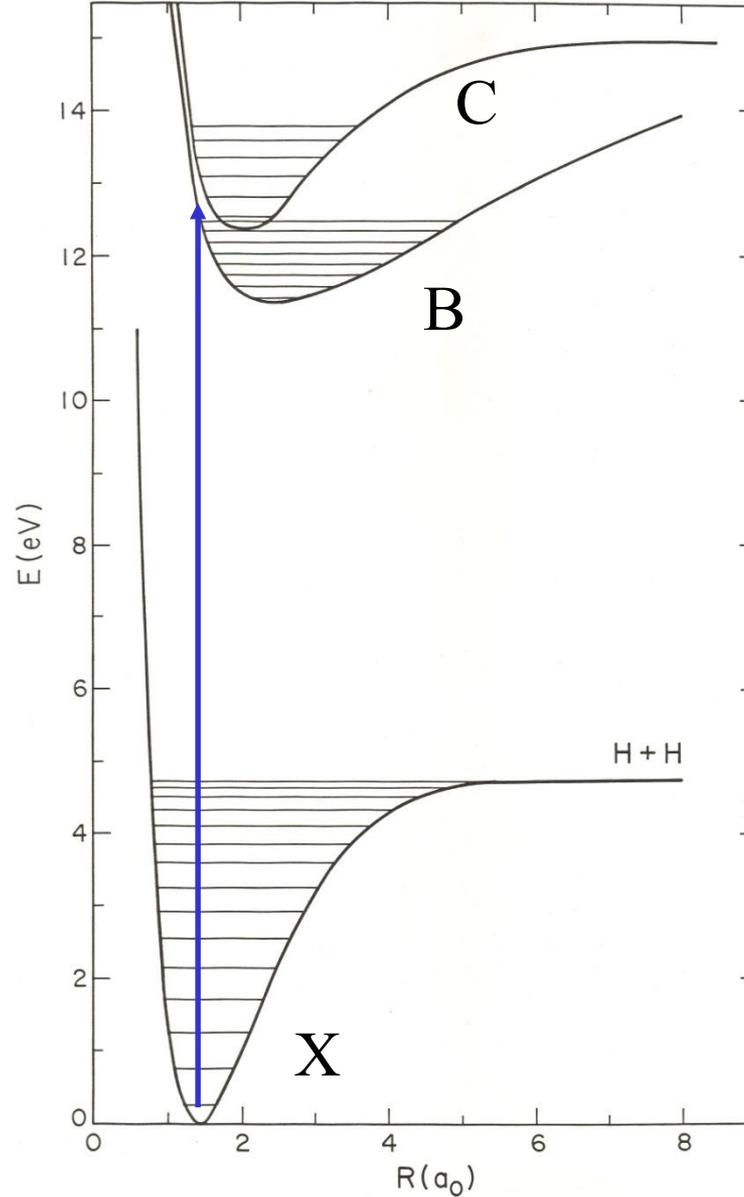
$\Delta J = -1$: P branch

R(0): $J_u = 1 - J_l = 0$

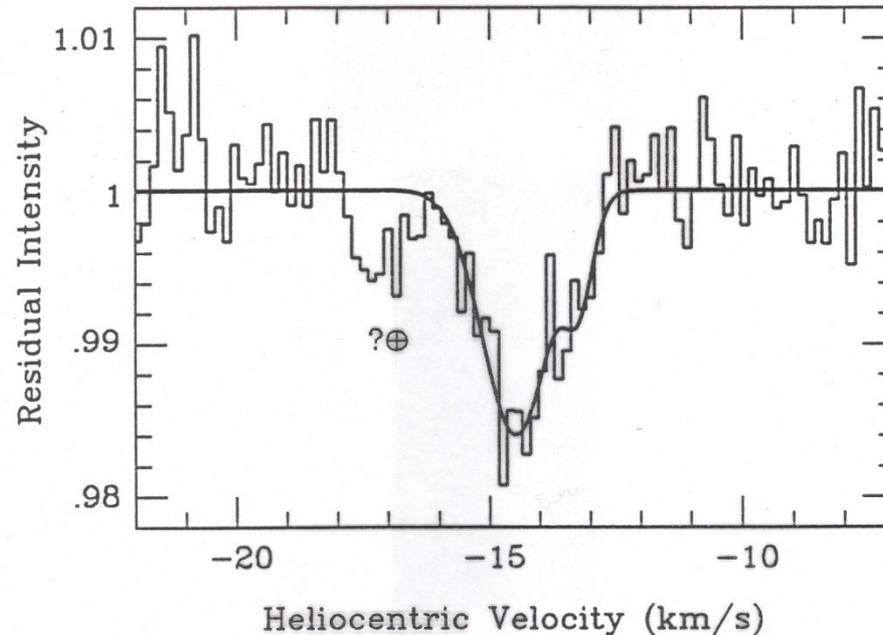
H₂ absorption

B-X: Lyman system
C-X: Werner system

P-lines: $J_u - J_l = -1$
R-lines: $J_u - J_l = 1$



Example: NH toward ζ Oph



Crawford & Williams 1997

- Ultra high resolution observations $R=800000$
- Two component fit:
 - $v=-14.5 \text{ km s}^{-1}$, $b=0.91\pm 0.15 \Rightarrow T < 750 \text{ K}$
 - $v=-13.2 \text{ km s}^{-1}$, $b=0.33\pm 0.13 \Rightarrow T < 100 \text{ K}$
- Upper limits on T from line width exclude $\text{N} + \text{H}_2 \rightarrow \text{NH} + \text{H}$ reaction

Lines are narrow (except CH^+) \rightarrow excludes strong shocks!

Typical abundances (ζ Oph, ζ Per)

- $\text{CH} \approx 2 \times 10^{13} \text{ cm}^{-2} \Rightarrow x(\text{CH}) = N(\text{CH})/N(\text{H}_2) \approx 4 \times 10^{-8}$
- $\text{OH} \approx 4 \times 10^{13} \Rightarrow x(\text{OH}) \approx 8 \times 10^{-8}$
- $\text{NH} \approx 9 \times 10^{11} \Rightarrow x(\text{NH}) \approx 2 \times 10^{-9}$
- $\text{CO} \approx 10^{14} - 10^{15} \Rightarrow x(\text{CO}) \approx 2 \times 10^{-7} - 2 \times 10^{-6}$
- $\text{CH}^+ \approx 10^{12} - 10^{13} \Rightarrow x(\text{CH}^+) \approx 2 \times 10^{-9} - 2 \times 10^{-8}$

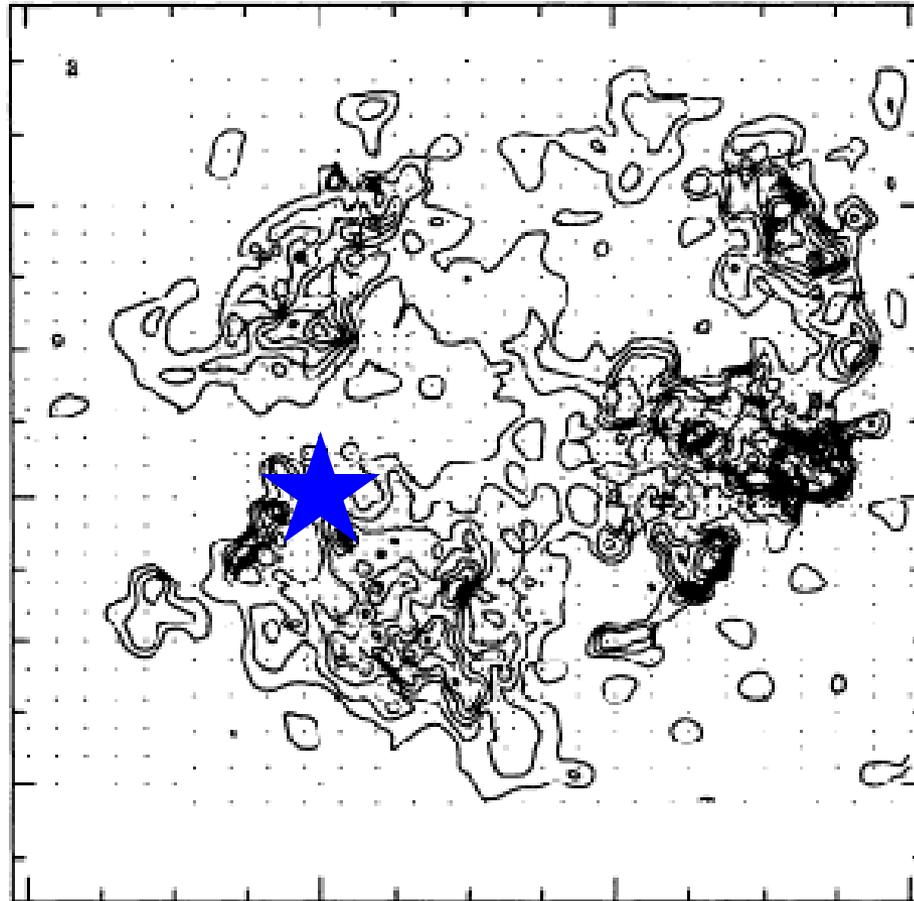
- The fact that the observed column densities of CH, OH \gg NH indicates that grain surface chemistry not dominant for carbon- and oxygen chemistry in diffuse clouds

See Appendix on how to derive N from observed equivalent width

Translucent clouds

- Can be isolated clouds, edges of dark clouds or GMCs, high-latitude clouds
- Visible absorption lines of CH^+ , CH , CN , C_2 ,
- UV absorption lines with FUSE \Rightarrow H_2 , HD , CO
 - Only for some clouds
- Clouds thick enough for mm emission \Rightarrow CO and very weak $[\text{C I}]$, CS , CN , HCO^+ , ***'CO dark gas'***
- Mm absorption lines against background quasars \Rightarrow ***surprisingly*** large column densities of HCO^+ , HCN , ...
- Infrared absorption \Rightarrow threshold for ice mantle formation at $A_V \geq 3$ mag

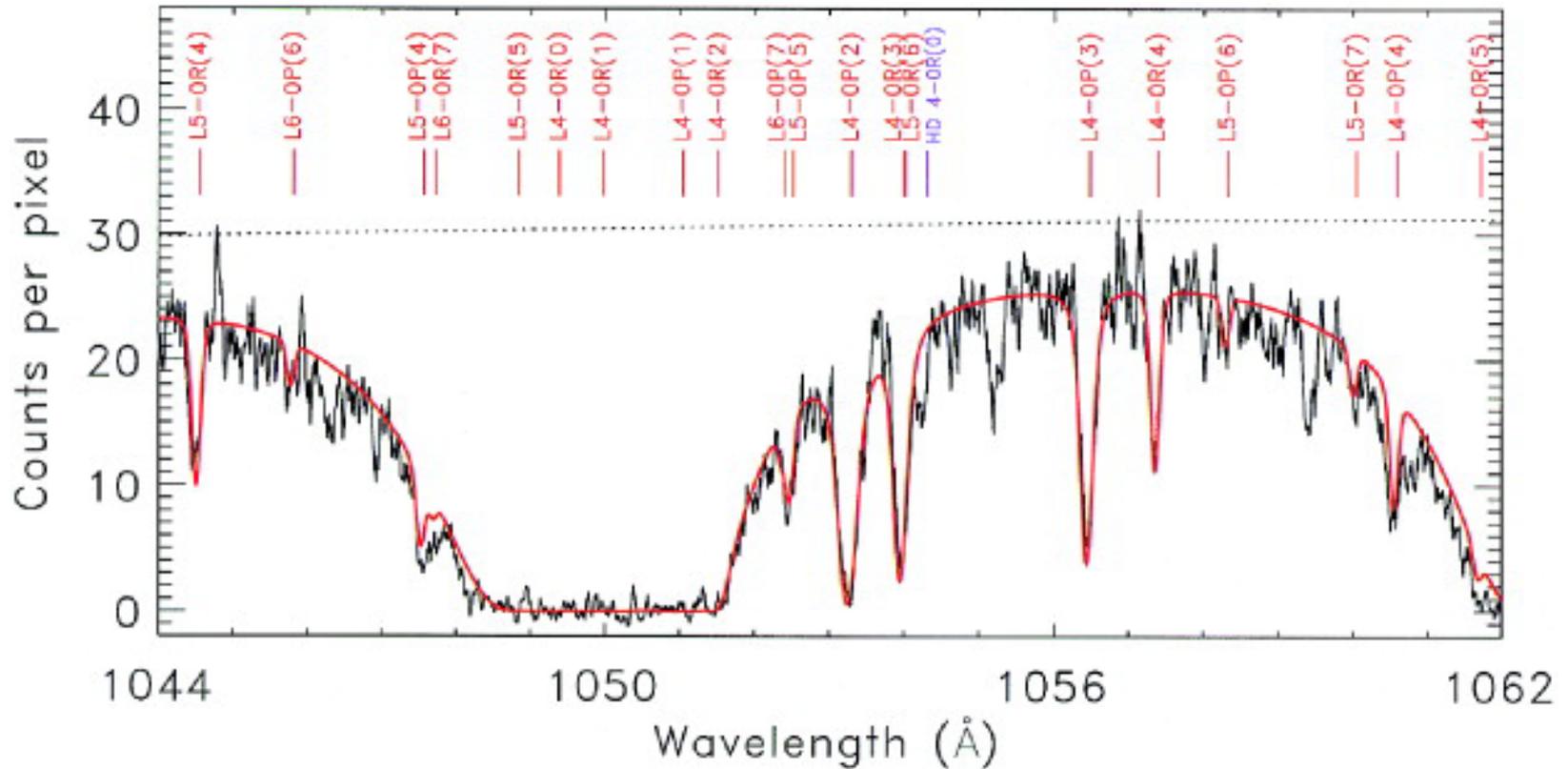
High latitude cloud toward HD 210121



CO $J=1-0$ map

UV spectroscopy: FUSE

HD 73882



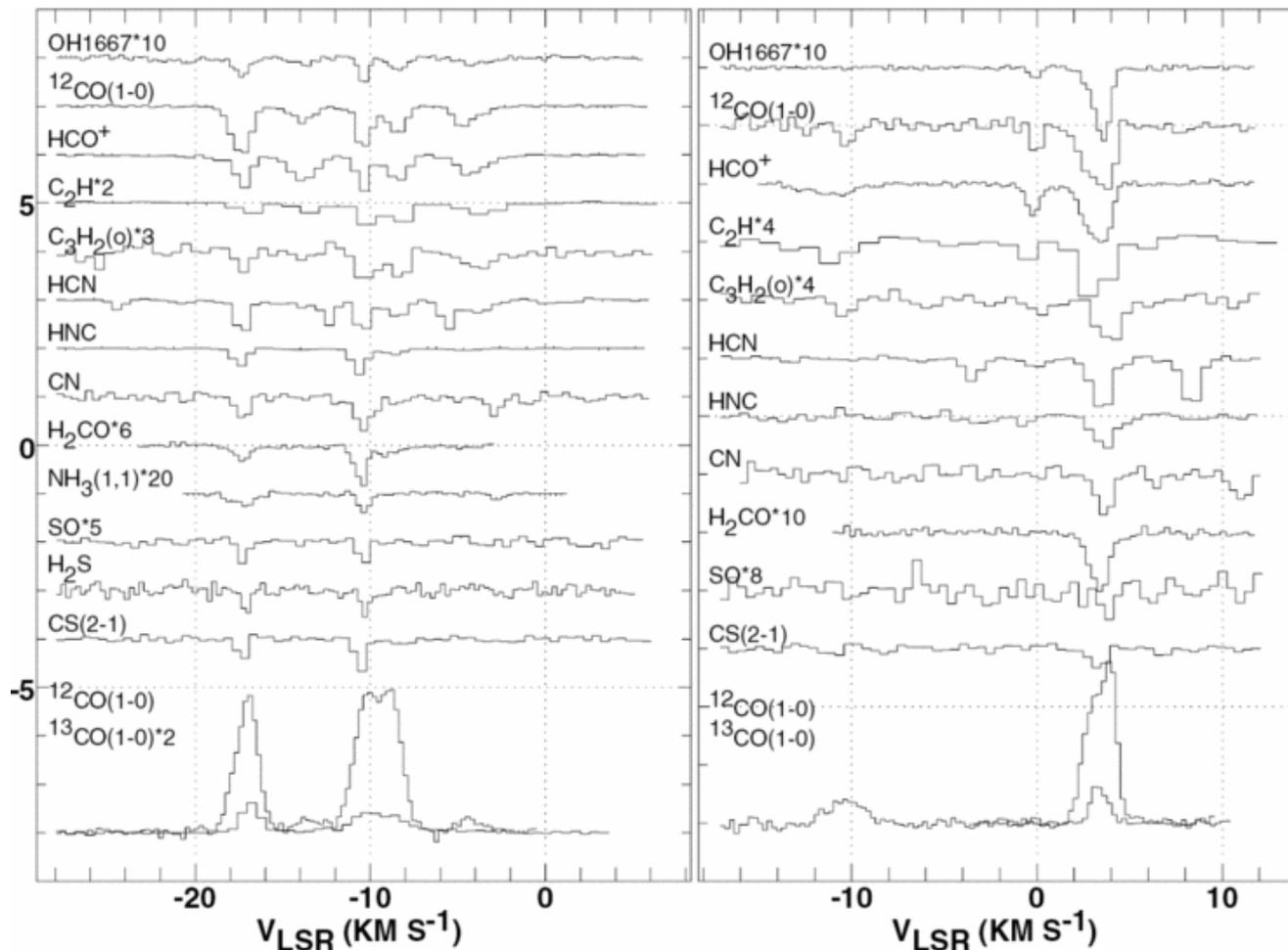
Snow et al. 2000

FUSE spectrum toward HD 73882 translucent cloud
showing H₂ and HD lines

Mm absorption lines: quasars

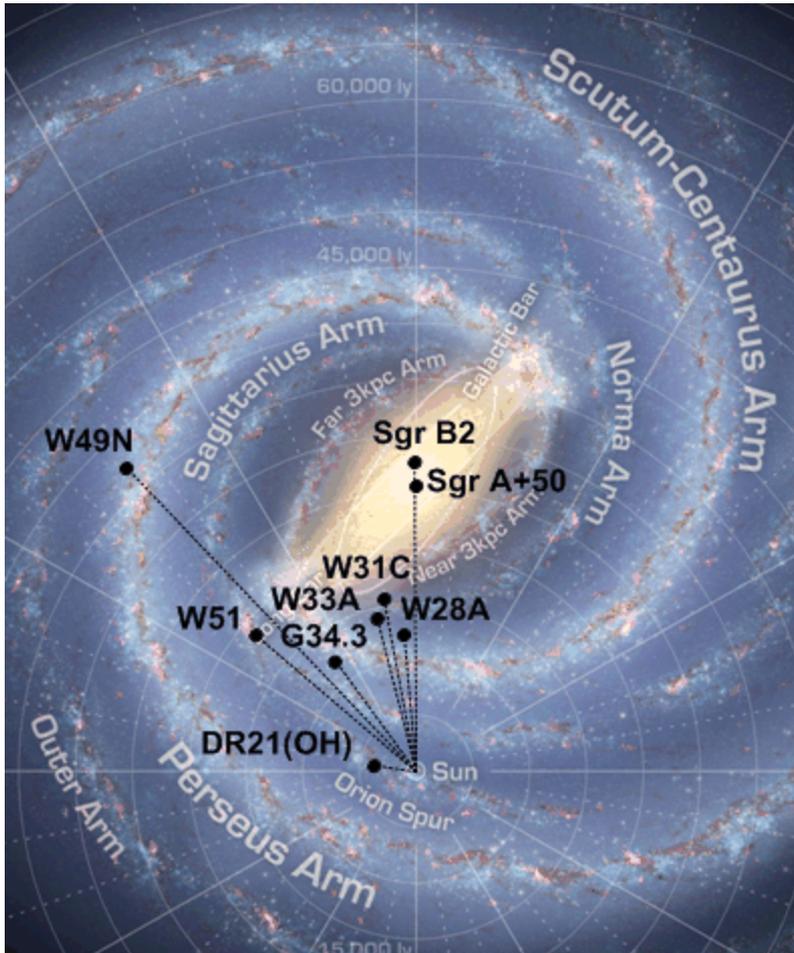
B0355+508

B0212+735



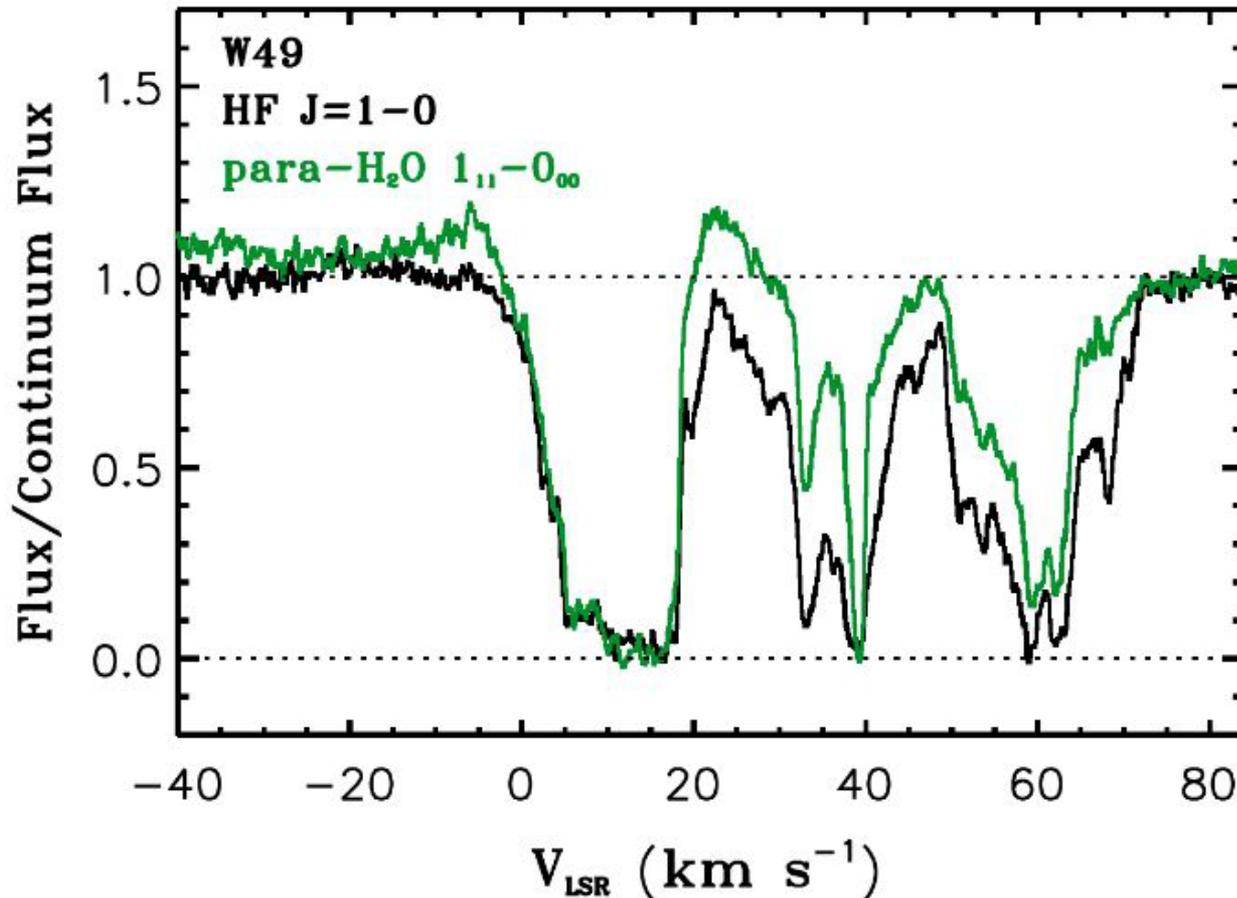
Lucas & Liszt 94-05
Hogerheijde et al. 1995
Liszt 2005, IAU 231

Far-infrared absorption



- Absorption against bright far-IR continuum
- Clouds $A_V \sim \text{few mag}$
- Hydrides have their lowest transitions at higher frequencies (submm/far-IR wavelengths) because of lower mass than CO
- All molecules in ground level \rightarrow simple analysis

Far-IR absorption lines: HF and H₂O



Herschel-
HIFI

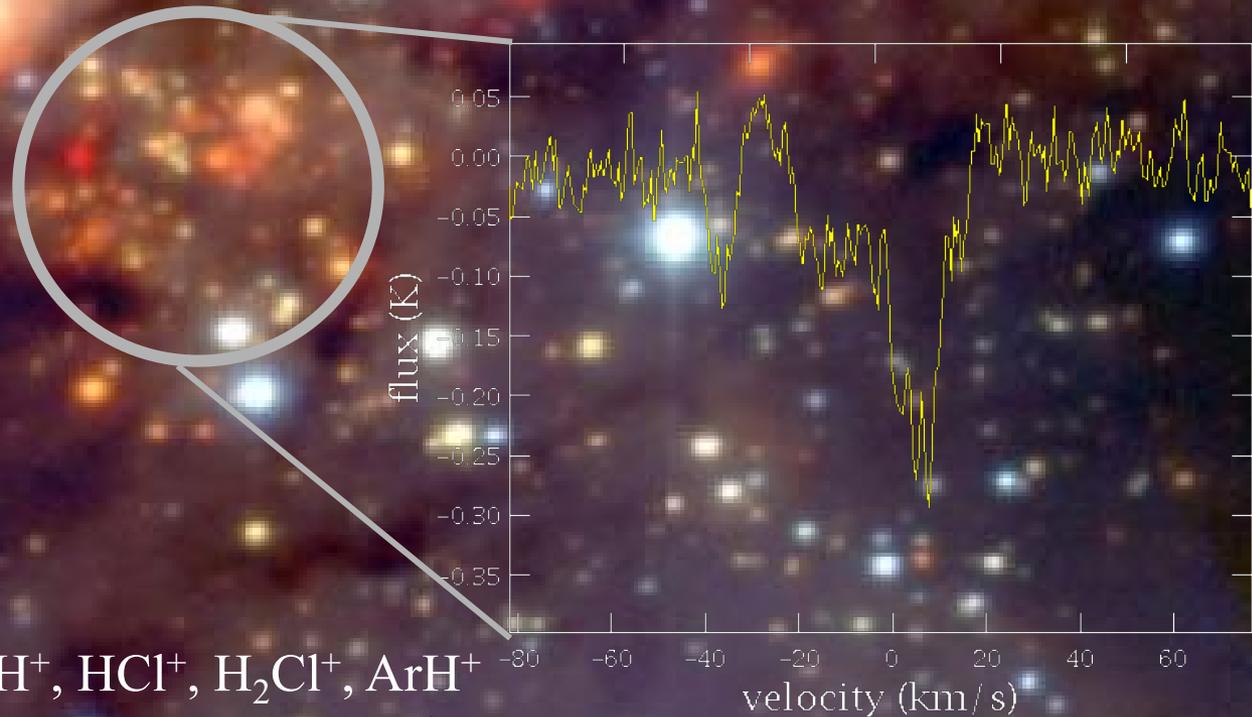
Gerin et al 2016

- Every feature is a diffuse or translucent cloud along the line of sight
- HF as tracer of H₂ column density because of simple chemistry
- Constant H₂O/H₂ abundance of 5×10^{-8}

Herschel Surprise: H_2O^+ widespread

Points to an ISM phase with high H/H_2 (see Exercise on ArH^+)

W3 IRS5



Benz, Bruderer et al. 2010
Gerin et al. 2010, Ossenkopf et al. 2010

Also observed: OH^+ , CH^+ , SH^+ , HCl^+ , H_2Cl^+ , ArH^+
 OH , H_2O , CH , NH , HF , SH , HCl

4.2 Physical structure: diagnostics

Species	Probes	Diffuse clouds	Transl. clouds
H_2 low J	T	+	-
H_2 high J , $v>0$	I_{UV} , formation	+	-
C_2 low J	T	+	+
C_2 high J	$I_{\text{R}}/n_{\text{H}}$	+	+
$\text{C}, \text{C}^+, \text{O}(J)$	n_{H}, T	+	-
CO low J	n_{H}, T	+	+
$\text{CN } J$	$n(\text{e})$	+	+
HCO^+, \dots	$n(\text{e}), n_{\text{H}}$	-	+

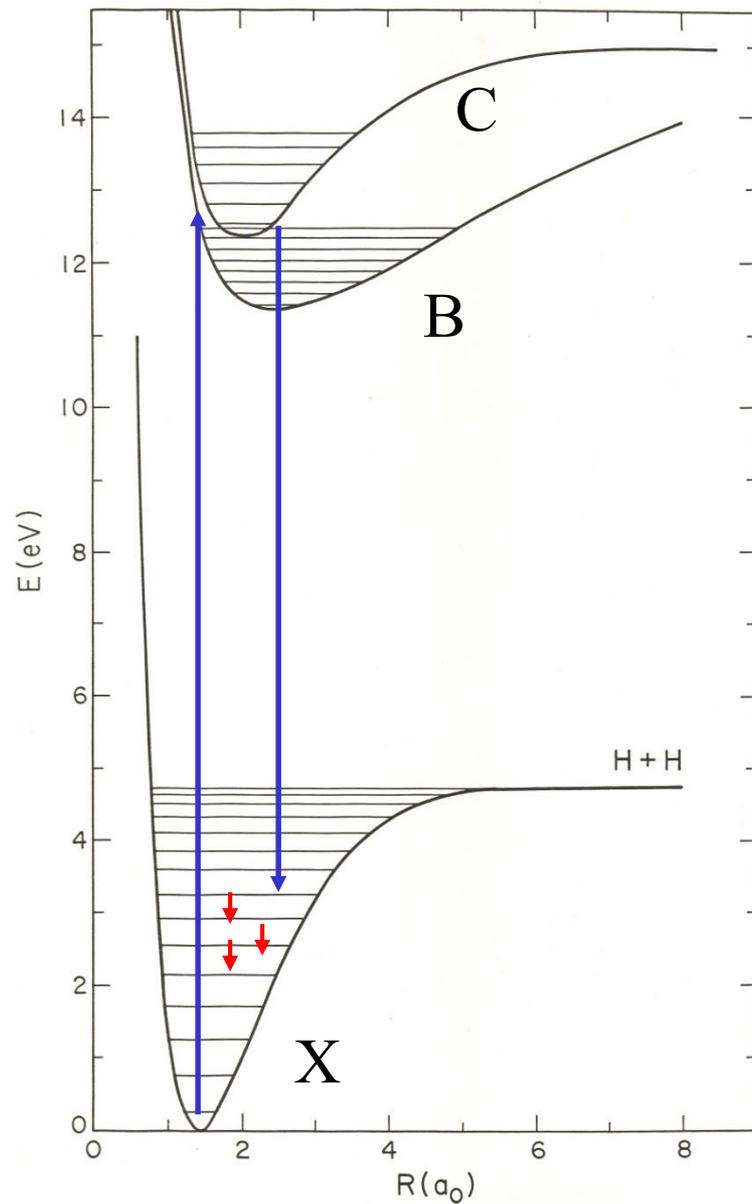
a. Rotational excitation of H₂

- H₂ lines out of $J = 0-7$ detected with Copernicus + FUSE
- Population distribution non-thermal
 - Low J : excitation dominated by collisions
⇒ *sensitive to T and n_H*
 - Abundance H⁺ large enough that ortho/para exchange rapid and $J=1/J=0$ gives T_{kin}
 - High J : energy levels lie very high (> 1000 K) ⇒ not populated by collisions at $T = 40 - 80$ K ⇒ populated by optical pumping process through B → X and C → X systems
⇒ *proportional to interstellar radiation field I_{UV}*
 - Formation process may play a small role as well

I_{UV} is scaling factor w.r.t. general interstellar radiation field (ISRF) in 6-13.6 eV range (also called G_0)

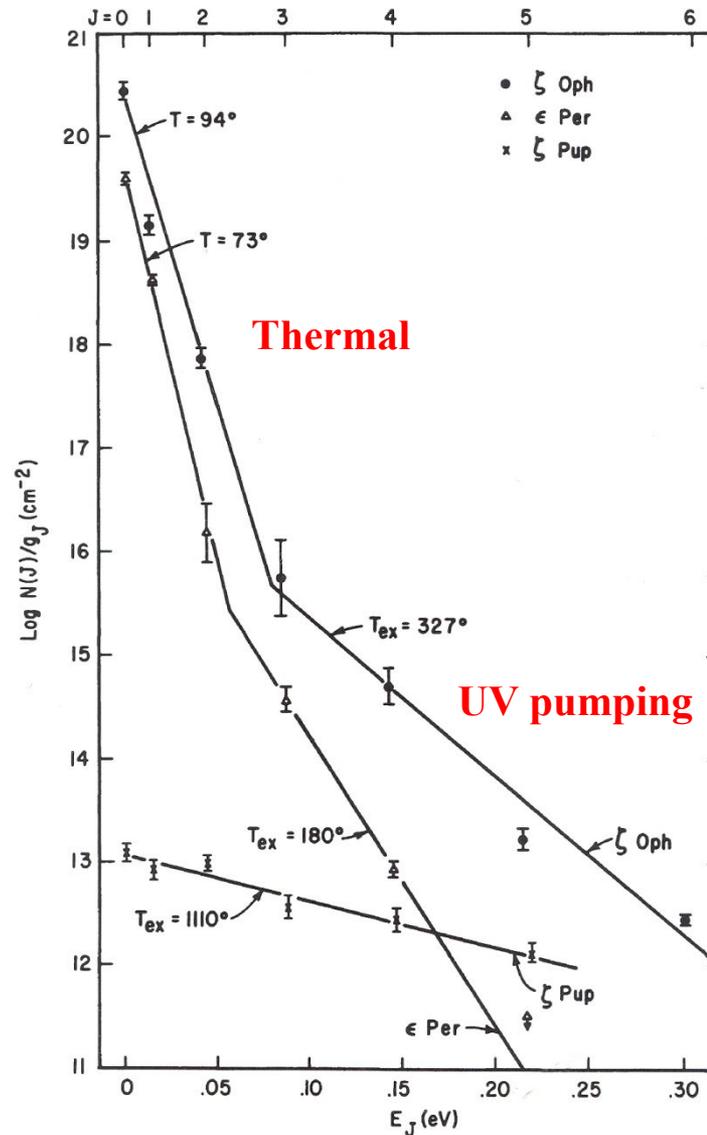
H₂ excitation

B-X: Lyman system
C-X: Werner system



Observed H₂ rotational excitation

$\ln N_J/g_J$

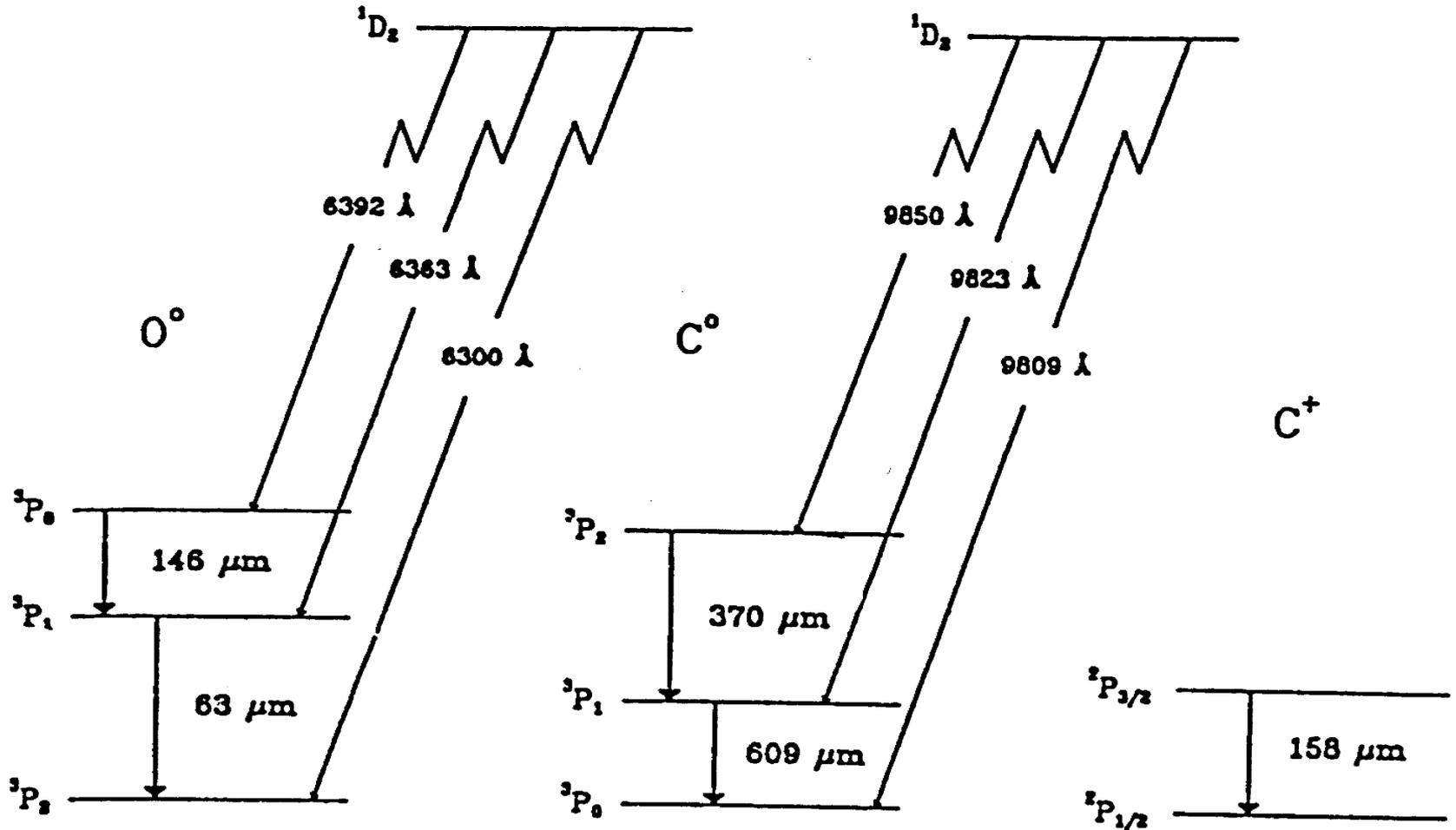


Diffuse clouds

b. Other diagnostics

- C_2 : similar to H_2 , but pumping by radiation around $1 \mu\text{m} \Rightarrow$ *sensitive to I_R , T and n_H*
- CO rotational excitation
 - Small dipole moment \Rightarrow lowest levels can be populated by collisions even at low densities \Rightarrow *sensitive to T and n_H*
- C, C^+ , O fine-structure excitation
 - Fine-structure populations determined by collisions \Rightarrow *sensitive to T and n_H*

O I, C I and C II fine structure lines



Results physical parameters

- Diffuse clouds

Edge

Center

- n_{H} : 100 → 500 cm⁻³ gradient $n_{\text{H}}=n(\text{H})+2n(\text{H}_2)$
- T : 100-200 → 25-40 K
- I_{UV} : 1-2×ISRF → decrease depending on λ

- Translucent clouds

- Similar, but with temperatures down to 15 K and densities up to a few thousand cm⁻³

*Bottom line: diffuse cloud densities a few hundred cm⁻³,
translucent cloud densities a few thousand cm⁻³*

4.3 Chemistry networks

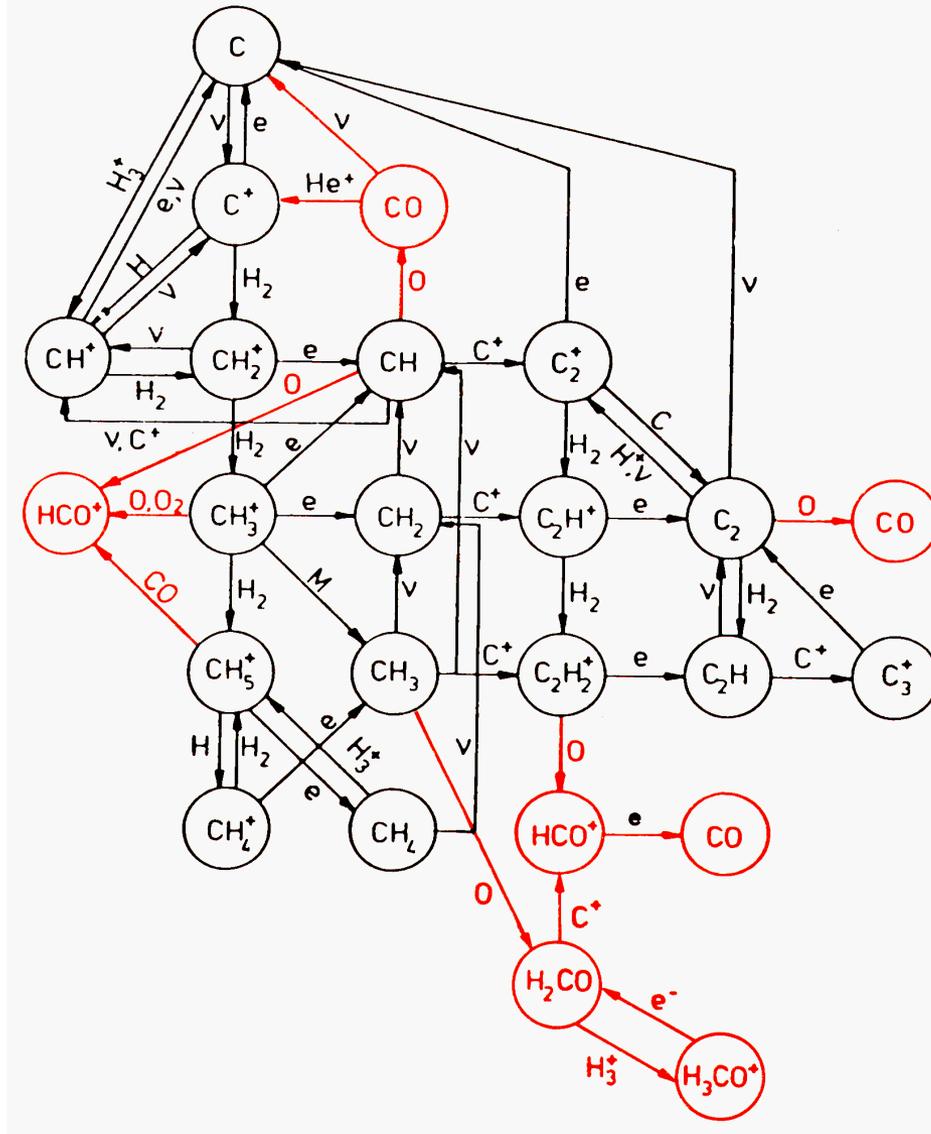
	k (cm ³ s ⁻¹)
■ <u>Formation</u> of bonds	
■ Radiative association:	$\sim 10^{-17} - 10^{-14}$
■ Grain surface:	$\sim 10^{-17}$
■ <u>Destruction</u> of bonds	
■ Photo-dissociation:	$\sim 10^{-10} - 10^{-8} \text{ s}^{-1}$
■ Dissociative recombination:	$\sim 10^{-7} - 10^{-6}$
■ <u>Rearrangement</u> of bonds	
■ Ion-molecule reactions:	$\sim 10^{-9} - 10^{-8}$
■ Neutral-neutral reactions:	$\sim 10^{-10} - 10^{-9}$

Hydrogen is dominant element \Rightarrow if species can react with H or H₂, it will be the dominant route

Carbon chemistry

- Need to have ions and molecules to start ion-molecule chemistry
- I.P. of C < 13.6 eV \Rightarrow carbon mostly C⁺
- $C^+ + H_2 \rightarrow CH_2^+ + h\nu$ possible at low T (initiating reaction)
- Once CH₂⁺ formed, rapid ion-molecule reactions lead to CH, C₂, ...
- $C^+ + H_2 \not\rightarrow CH^+ + H$: endothermic by 0.4 eV

Carbon chemistry and its coupling with oxygen



Vertical

Down: H_2

Up: photon ν

Horizontal

Ion: e^-

Neutral: C^+

Oxygen chemistry

- I.P. of O > 13.6 eV \Rightarrow oxygen mostly neutral O
- Ionization provided by cosmic rays
 - H_2 or $\text{H} + \text{C.R.} \rightarrow \text{H}_2^+$ or $\text{H}^+ + \text{C.R.} + e$
 - $\text{H}_2^+ + \text{H}_2 \rightarrow \text{H}_3^+ + \text{H}$ (very fast)
- H^+ or H_3^+ can react with oxygen
 - $\text{H}^+ + \text{O} \leftrightarrow \text{H} + \text{O}^+ \quad \Delta E = 227 \text{ K}; \text{O}^+ + \text{H}_2 \rightarrow \text{OH}^+ + \text{H}$
 - $\text{H}_3^+ + \text{O} \rightarrow \text{OH}^+ + \text{H}_2$

Oxygen chemistry

- Once OH^+ formed, rapid ion-molecule reactions lead to OH, H_2O and CO
- Note that OH abundance proportional to cosmic ray ionization rate $\zeta_{\text{CR}} \Rightarrow$ can use observed OH abundance to determine ζ_{CR}

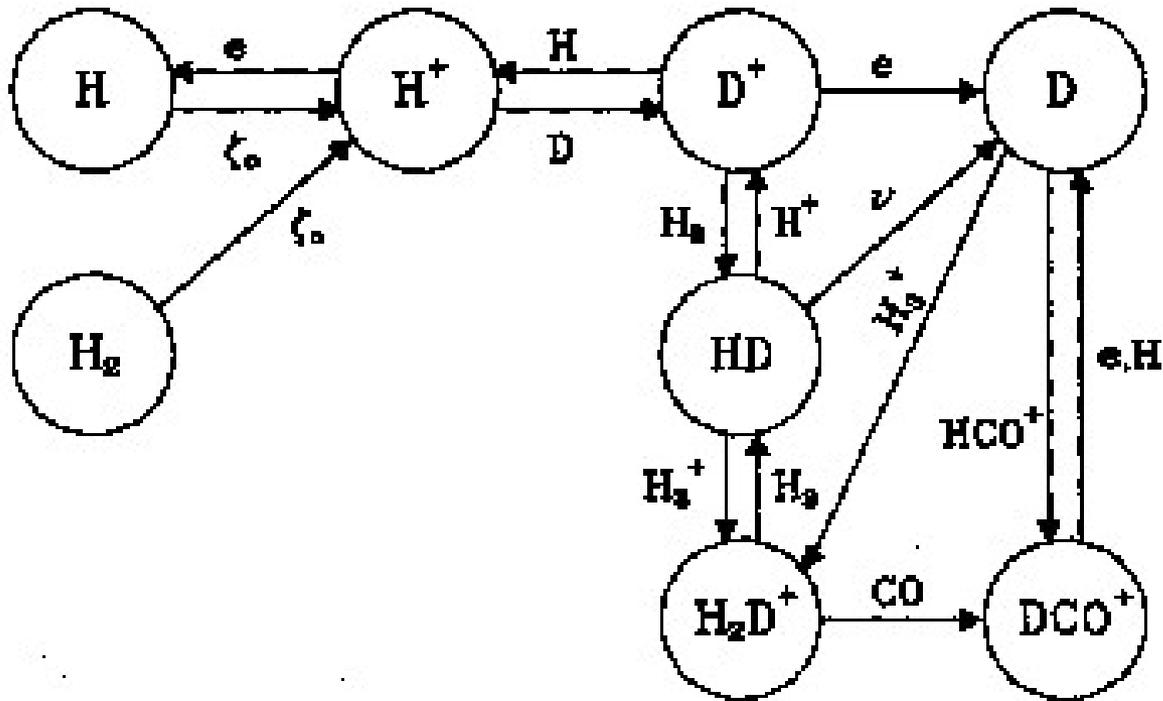
Nitrogen chemistry

- I.P. N > 13.6 eV => nitrogen mostly neutral N
- Nitrogen chemistry initiated by
 - $\text{N} + \text{H}_3^+ \rightarrow \text{NH}_2^+ + \text{H}$? energy barrier?
 - $\text{N}^+ + \text{H}_2 \rightarrow \text{NH}^+ + \text{H}$? small barrier ~ 100 K
- Link with carbon chemistry
 - $\text{CH}, \text{C}_2 + \text{N} \rightarrow \text{CN} + \text{H}, \text{C}$
 - $\text{CH}_3^+ + \text{N} \rightarrow \text{H}_2\text{CN}^+ + \text{H}$
 - $\text{H}_2\text{CN}^+ + e \rightarrow \text{HCN} \text{ or } \text{HNC} + \text{H}$

Deuterium chemistry

- HD formed mostly by rapid gas-phase reactions (*not on grains*), in contrast with H₂
 - $\text{H}^+ + \text{D} \leftrightarrow \text{H} + \text{D}^+$
 - $\text{D}^+ + \text{H}_2 \rightarrow \text{HD} + \text{H}^+$
 - HD more rapidly destroyed by photodissociation than H₂, because column densities too small for self-shielding
- ⇒ Both formation and destruction up to 10⁵-10⁶ times faster;
Result: overall HD/H₂ ~ 10⁻⁶-10⁻⁵ vs [D]/[H] = 1.5 × 10⁻⁵

Deuterium network



4.4 Steady-state models

- Because photorates change with depth into cloud, models of diffuse clouds need to be depth-dependent
- Most reactions are fast (except H₂ formation) ⇒ steady-state usually good assumption: $dn_i/dt=0$

density $n_i(z) = \frac{\text{formation rate}}{\text{destruction rate}} \quad \text{cm}^{-3}$

column density $N_i = \int n_i(z) dz \quad \text{cm}^{-2}$

Parameters

- Geometry: plane parallel
- Elemental abundances: C, O, N, S, Fe, Mg, ...
 - Often constrained by observations
- Density structure: $n_{\text{H}}(z)$
 - Constrained by observations
- Temperature structure: $T(z)$
 - Constrained by observations or by solving thermal balance
- Incident radiation field: $I_{\text{UV}} \times$ standard field, plus grain absorption + scattering parameters
 - Constrained by observations
 - $I_{\text{V}} = I_0 10^{-0.4A_{\text{V}}}$ (definition of A_{V})
 - $I_{\text{UV}}(\lambda) = I_0 \exp(-\tau_{\text{d}}(\lambda))$ $\tau_{\text{d}}(\lambda) = n_{\text{d}} C_{\text{ext}}(\lambda) L$
- Cosmic ray ionization rate: ζ (s^{-1})

Rate equations

$$\frac{dn(i)}{dt} = \sum_l \sum_j K_{lj} n(l) n(j) - n(i) \sum_j K_{ij} n(j) - R_{\text{acc}}(i) + t_{\text{evap}}(i)^{-1} n_s(i),$$

Gas

$$\frac{dn_s(i)}{dt} = \sum_l \sum_j k_{lj} n_s(l) n_s(j) - n_s(i) \sum_j k_{ij} n_s(j) + R_{\text{acc}}(i) - t_{\text{evap}}(i)^{-1} n_s(i),$$

Surface

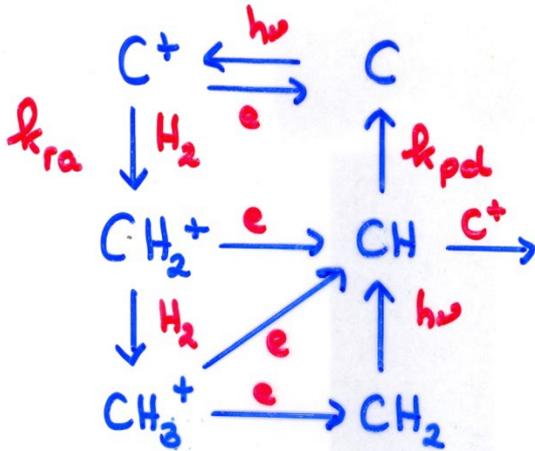
Databases (KIDA, UMIST, ...) parametrize rate coefficients

$k = \alpha (T/300)^\beta \exp(-\gamma/T) \text{ cm}^3 \text{ s}^{-1}$ for bimolecular reaction

$k = \alpha \text{ s}^{-1}$ for cosmic ray ionization

$k = \alpha \exp(-\gamma A_V) \text{ s}^{-1}$ for photodissociation or photoionization

Example: CH



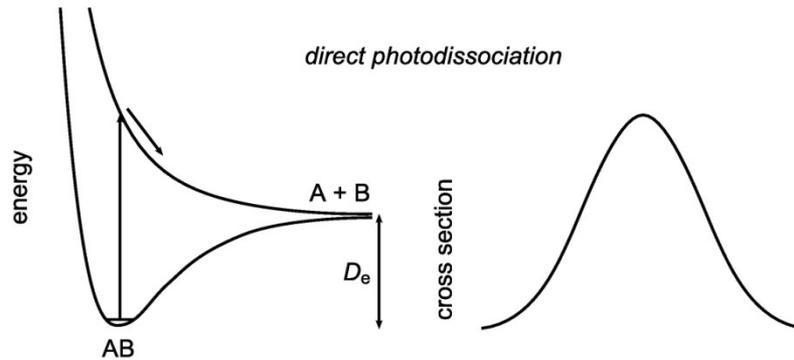
$$n(\text{CH}) \propto \frac{n(\text{C}^+)n(\text{H}_2)k_{ra}}{k_{pd}}$$

- All rates well known (within factor 2) \Rightarrow it is possible to fit observed CH column densities
 - cannot claim that all CH is formed by this scheme; up to 50% could result from other processes
- Test: can consistent model be made that reproduces all observational data with same set of physical conditions and molecular parameters?

Comprehensive models

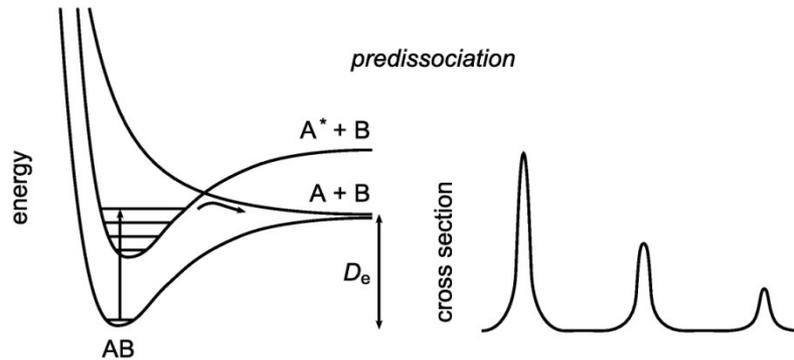
- Comprehensive models attempted by several groups, e.g., Black & Dalgarno (1977), van Dishoeck & Black (1986), Meudon group, UMIST,
- Chemistry governed by $\text{H} \rightarrow \text{H}_2$ and $\text{C}^+ \rightarrow \text{C} \rightarrow \text{CO}$ transitions
- Because of self-shielding, these transitions are *sharp* as functions of depth into cloud

Recall photodissociation Lecture 1



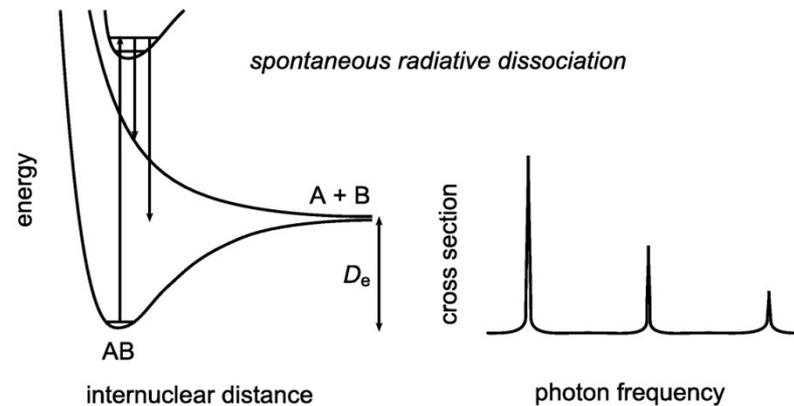
Direct p.d.

Ex: H_2^+ , OH, H_2O



Predissociation

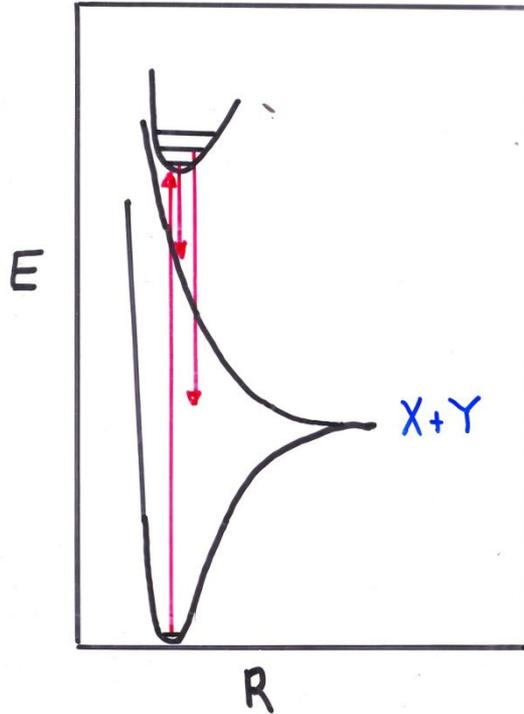
Ex: CO



**Spontaneous
radiative
dissociation**

Ex: H_2

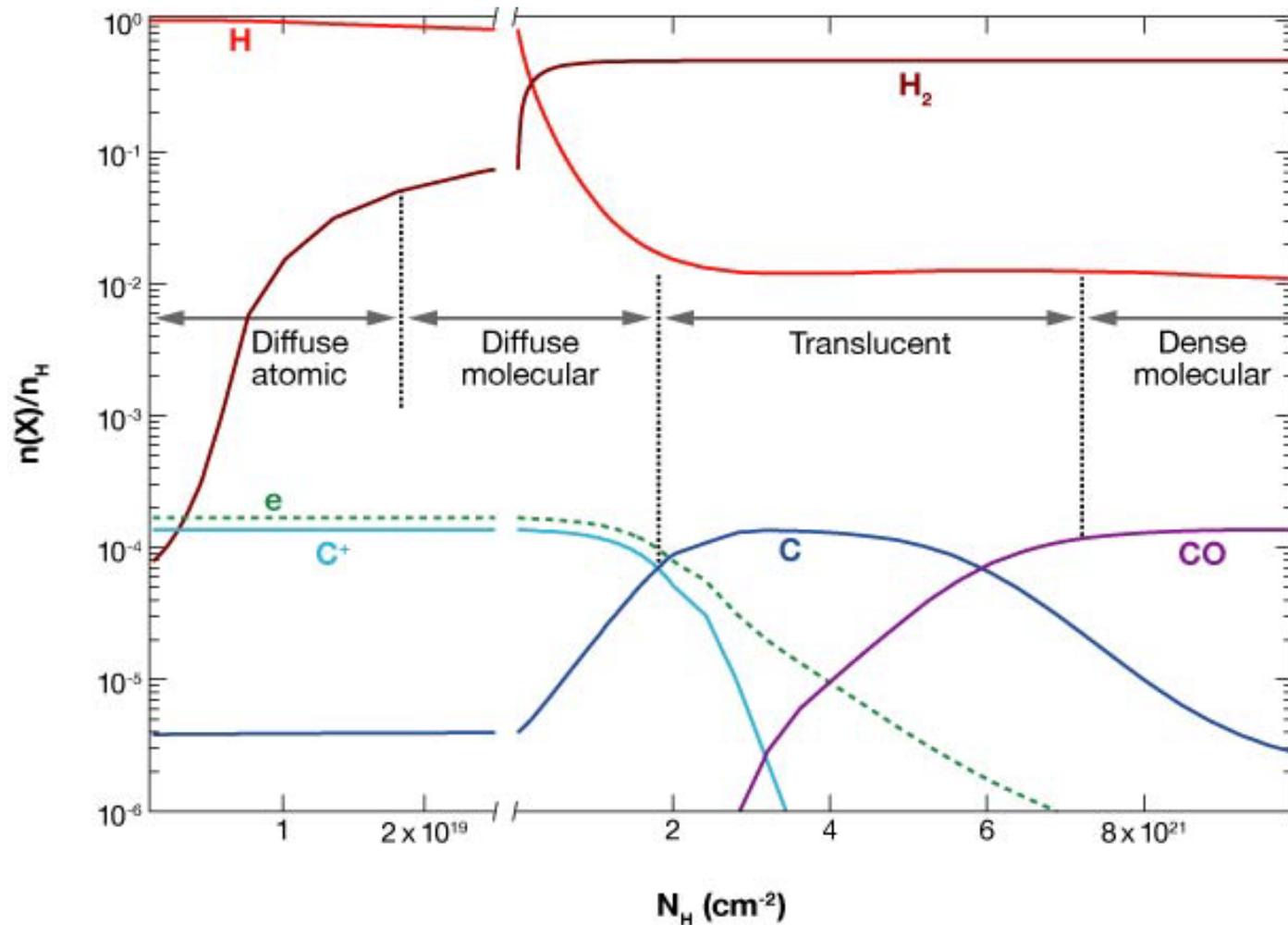
Photodissociation: self-shielding



Spontaneous radiation dissociation
Ex: H_2

- Photodissociation of both H_2 and CO occurs by line absorption \rightarrow lines can become optically thick \rightarrow **self-shielding** inside clouds
 - Molecules lying at the edge of the cloud absorb all available photons at that wavelength, so that molecules lying deeper in the cloud are protected: *sharp* transition!

Types of molecular clouds



 Snow TP, McCall BJ. 2006.

Annu. Rev. Astron. Astrophys. 44:367–414

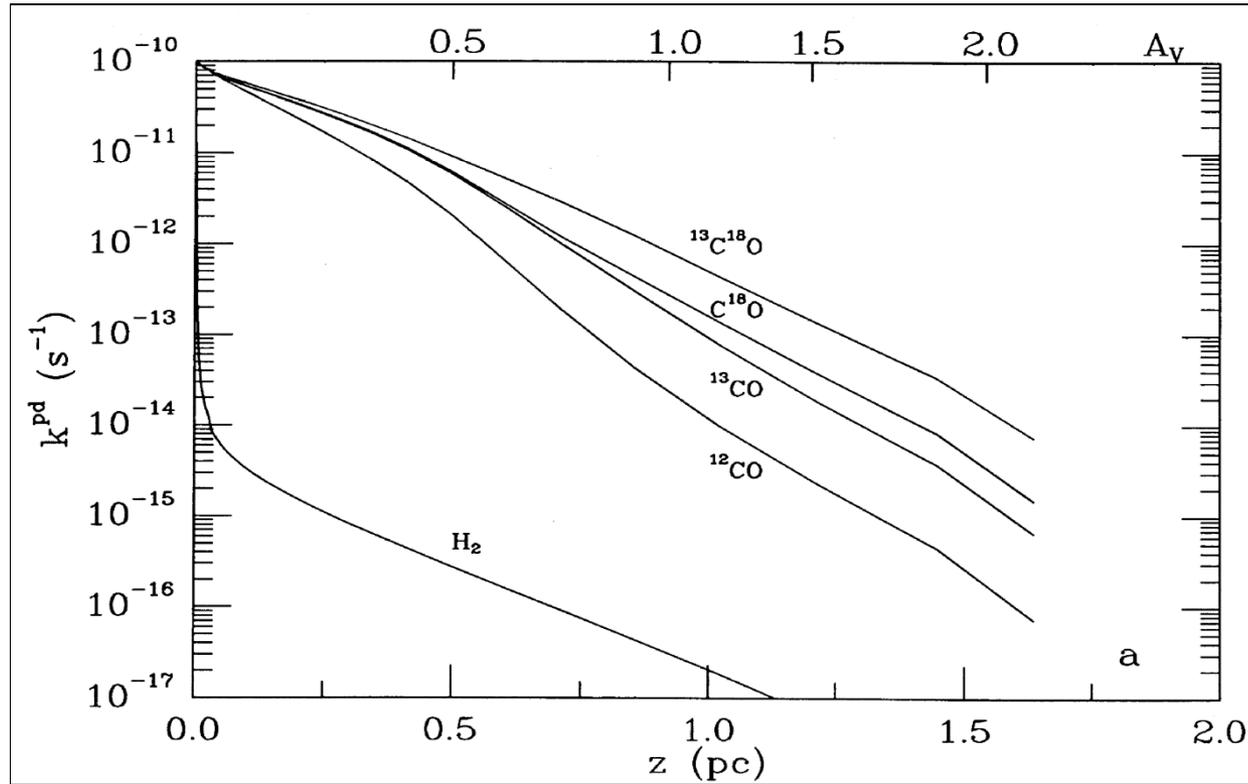
CO formation and destruction

- CO is most abundant molecule after H₂ and is easily observed through (sub-) mm lines ⇒
 - Good tracer of H₂
- At edge of cloud, most of carbon is C⁺
- Transition C⁺ → C → CO with increasing depth
- CO is very stable ($D_e = 11.09 \text{ eV} \Leftrightarrow 1118 \text{ \AA}$) ⇒ can only be dissociated at $912 \text{ \AA} < \lambda < 1118 \text{ \AA}$

CO photodissociation

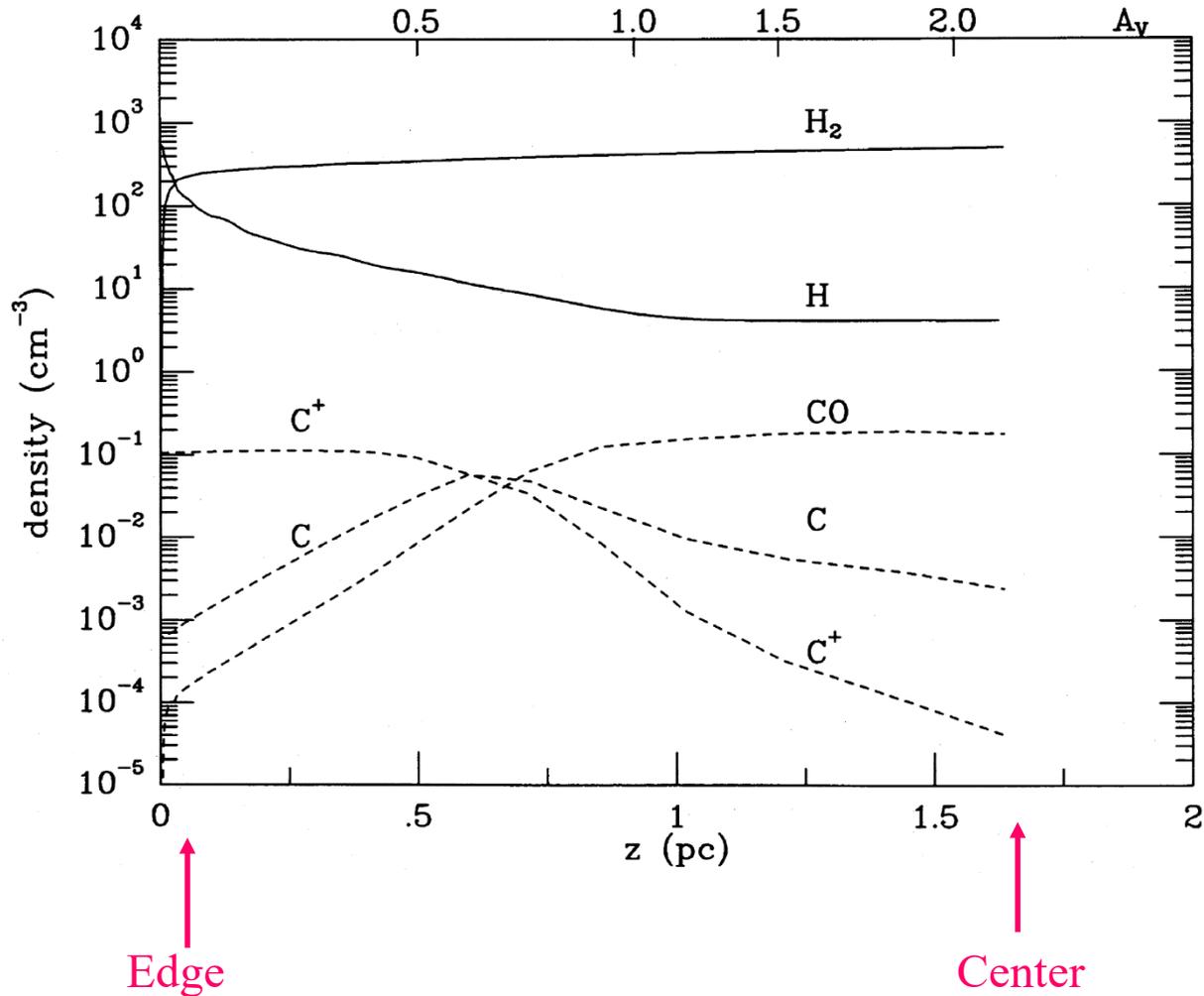
- Like H₂, CO has no direct dissociation channels \Rightarrow dissociation through line absorption (predissociation) \Rightarrow
 - Self-shielding, but at greater depth than H₂ because of lower abundance
 - Mutual shielding of CO by H₂
 - Mutual shielding of ¹³CO by ¹²CO
 - Attenuation by dust \Rightarrow complicated UV radiative transfer!
- Note that H₂ and CO photodissociation and C photoionization all occur in the same λ range 912-1100 Å
 - Important to know shape UV radiation field at short λ
- At $A_V \approx 1-2$ mag, CO / H₂ increases from 10^{-7} to 10^{-4}

Self-shielding of CO and H₂ Photodissociation rates



- Decrease of CO p.d. rate with depth is less steep than that of H_2 because of larger competition with dust for UV photons
- CO self-shielding starts at $N(\text{CO}) \approx 10^{15} \text{ cm}^{-2}$

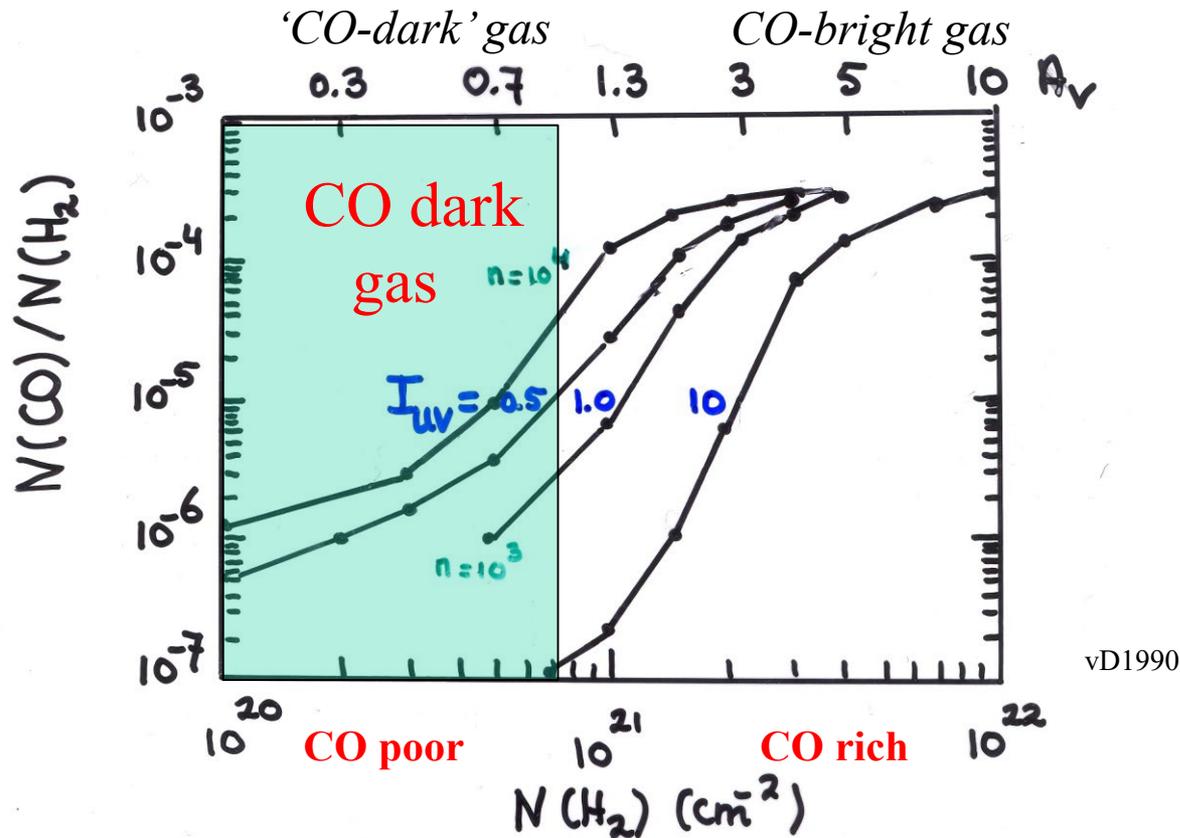
Densities of major species in translucent cloud



$T=15$ K
 $n_{\text{H}}=1500$ cm^{-3}
 $I_{\text{UV}}=1$

- Only column densities are observed \Rightarrow integrate over z

Column densities with N_{H} or A_{V}

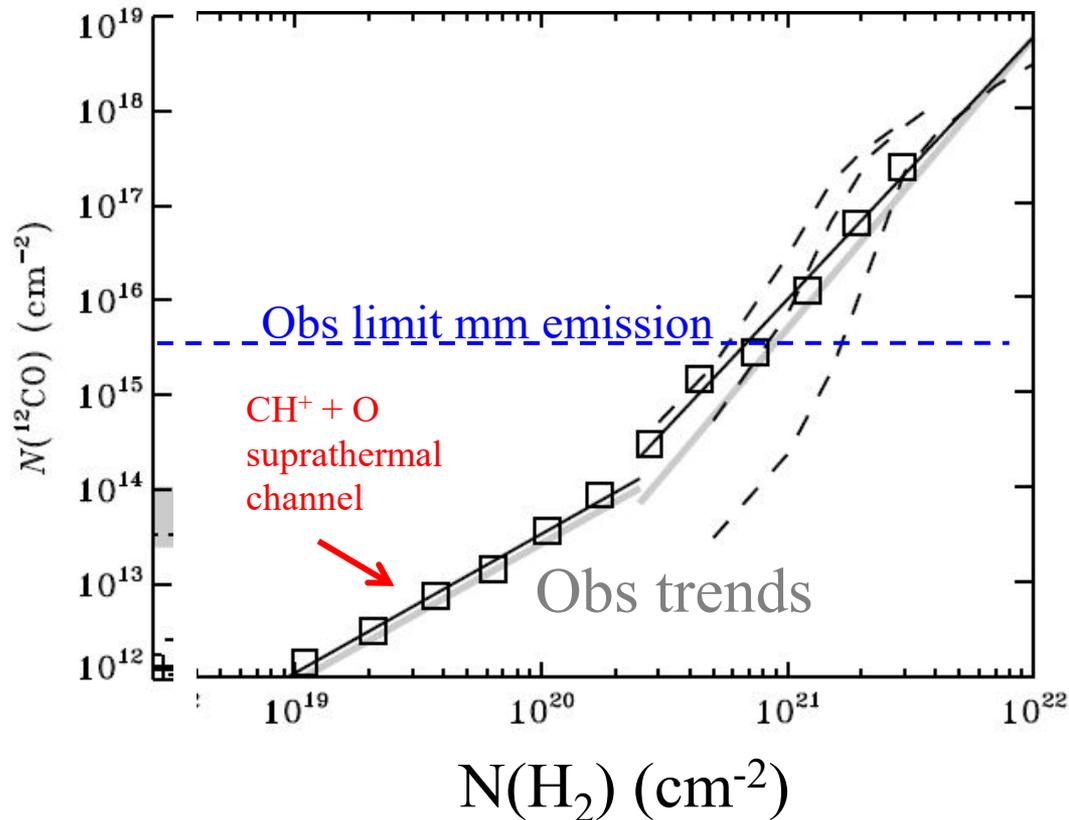


vD1990

⇒ Increase in CO/H_2 at $A_{\text{V}}=1-2$ mag from 10^{-7} to 10^{-4}

- Exact location and sharpness transition depend on
 - I_{UV} , n_{H} , elemental abundances
 - Presence of PAHs, chemistry details (S), ζ , grain details, geometry....

CO vs H₂ column density



- At low A_V , formation of CH^+ through superthermal chemistry of $\text{C}^+ + \text{H}_2$ needs to be invoked to fit observations CO in diffuse clouds

Successes models

- Column densities CH, C₂, C₃, CN, OH, HCl
O.K. within factor of 2
- Constraints on ζ and [D]/[H]
 - $\zeta=(7\pm3)\times 10^{-17} \text{ s}^{-1}$ (but see below H₃⁺, OH⁺)
 - [D]/[H] $\approx(1-2)\times 10^{-5}$

Failures models

- CH^+ too low by a factor of 100
- NH too low by factor of 20
- Polyatomic molecules HCO^+ , HCN , CH_3 ,
... too low

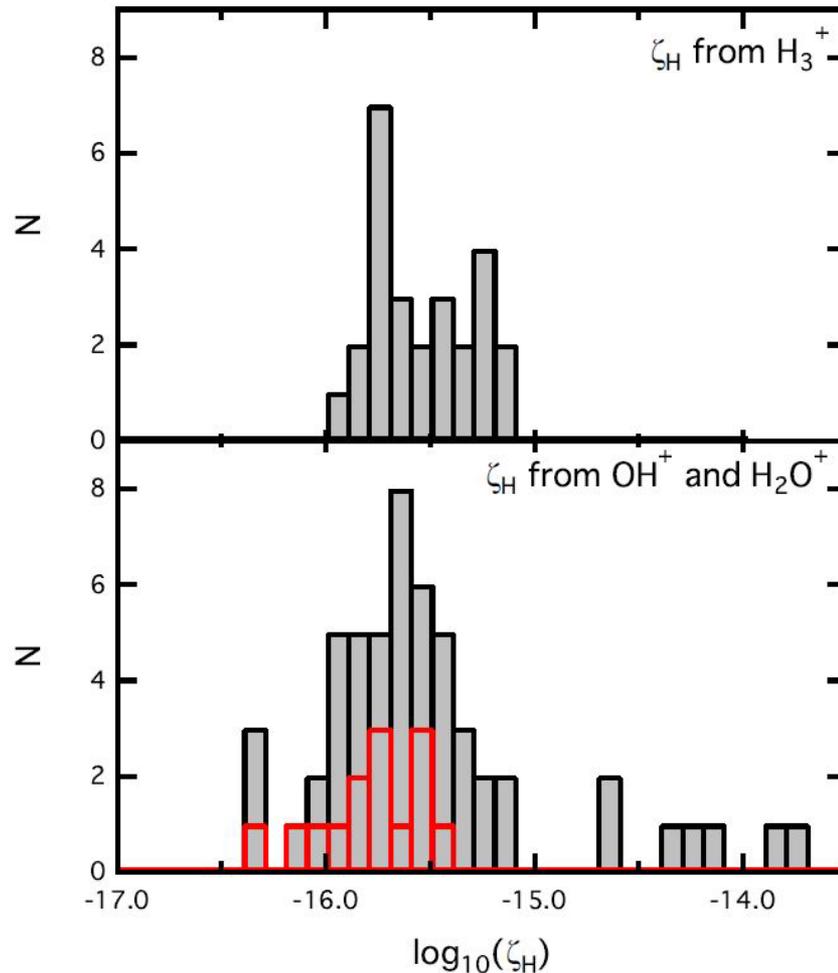
Possible remedies

- Weak shocks or turbulence
 - CH^+ : TDR models Godard et al. 2016
 - see also Chapter 5
 - HCO^+ , HCN , CO , CH_3 enhanced if CH^+ enhanced
- Surface chemistry
 - NH , CH_3 ?
- Cloud structure: clumpy, high H/H_2 , ...
 - CO
 - OH^+ , H_2O^+ , ArH^+

H_3^+ as probe of ζ

- Large column densities of $\sim 10^{14} \text{ cm}^{-2}$ found in several diffuse clouds (Galactic Center, ζ Per, others)
- $n(\text{H}_3^+) = (\zeta/k_{\text{DR}})n(\text{H}_2)/n(e)$
- $N(\text{H}_3^+) = n(\text{H}_3^+)L$ $L = \text{pathlength}$
- $n(e)/n(\text{H}_2) \approx N(e)/N(\text{H}_2) \approx N(\text{C}^+)/N(\text{H}_2) \approx 3.8 \times 10^{-4}$ **observed ζ Per**
- $k_{\text{DR}} = 2.6 \times 10^{-7} \text{ cm}^3 \text{ s}^{-1}$ **from exp**
- $N_{\text{H}} = 1.6 \times 10^{21} \text{ cm}^{-2}$, $\langle n_{\text{H}} \rangle = 250 \text{ cm}^{-3} \Rightarrow L = 2.1 \text{ pc}$
 \uparrow **obs** \uparrow **obs**
- $\Rightarrow \zeta = 1 \times 10^{-15} \text{ s}^{-1}$, factor of 10-20 larger than ‘canonical value’
- Survey gives range of 10^{-16} - 10^{-15} s^{-1} for number of sources

Cosmic ray ionization rate



Indriolo et al. 2015
Neufeld et al. 2017

Use both H_3^+ and OH^+ , H_2O^+ as probes

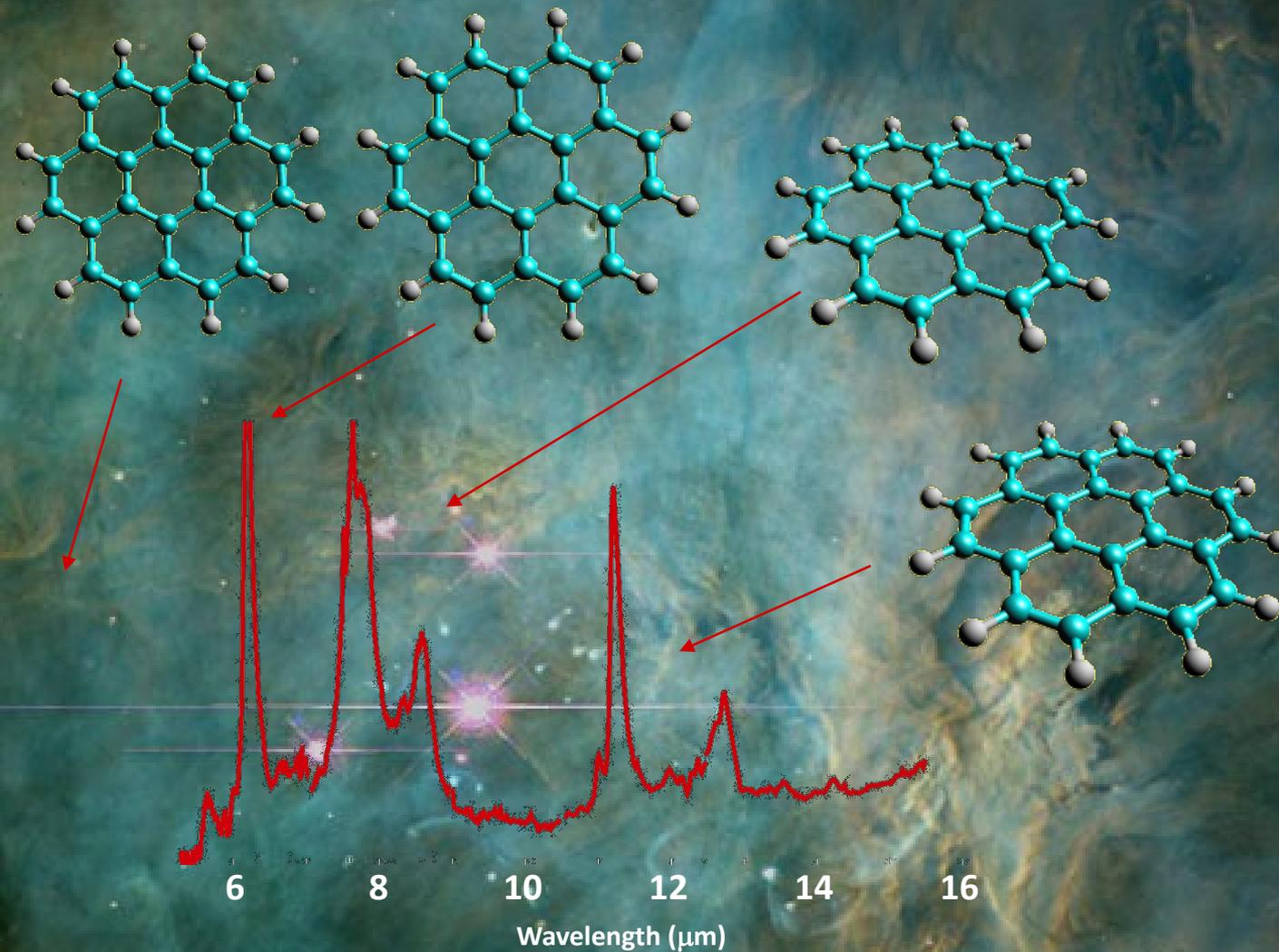
4.4 Photon-dominated regions (PDRs)

Tielens Chap 11

Hollenbach & Tielens 1997

- Diffuse and translucent clouds are examples of PDRs, i.e., clouds in which UV photons control the physical and chemical state of the cloud
- Traditionally, PDRs are dense molecular clouds located close to an OB star, in which the UV radiation field is enhanced by a factor of 10^5 w.r.t. average interstellar radiation field (ISRF)
 - Example: Orion Bar
- PDRs show very strong atomic fine-structure lines
 - E.g. [C II] 158 μm , [C I] 610 μm , [O I] 63 μm
- And submillimeter lines of molecules
 - E.g. CO 7-6, HCO⁺ 4-3
- And strong H₂ mid- and near-IR lines and PAH features

PAH vibrational modes



“Unidentified infrared emission bands”

PAHs heat gas through photoelectric effect

Horsehead nebula as a PDR



The Horsehead Nebula
(VLT KUEYEN + FOR S 2)

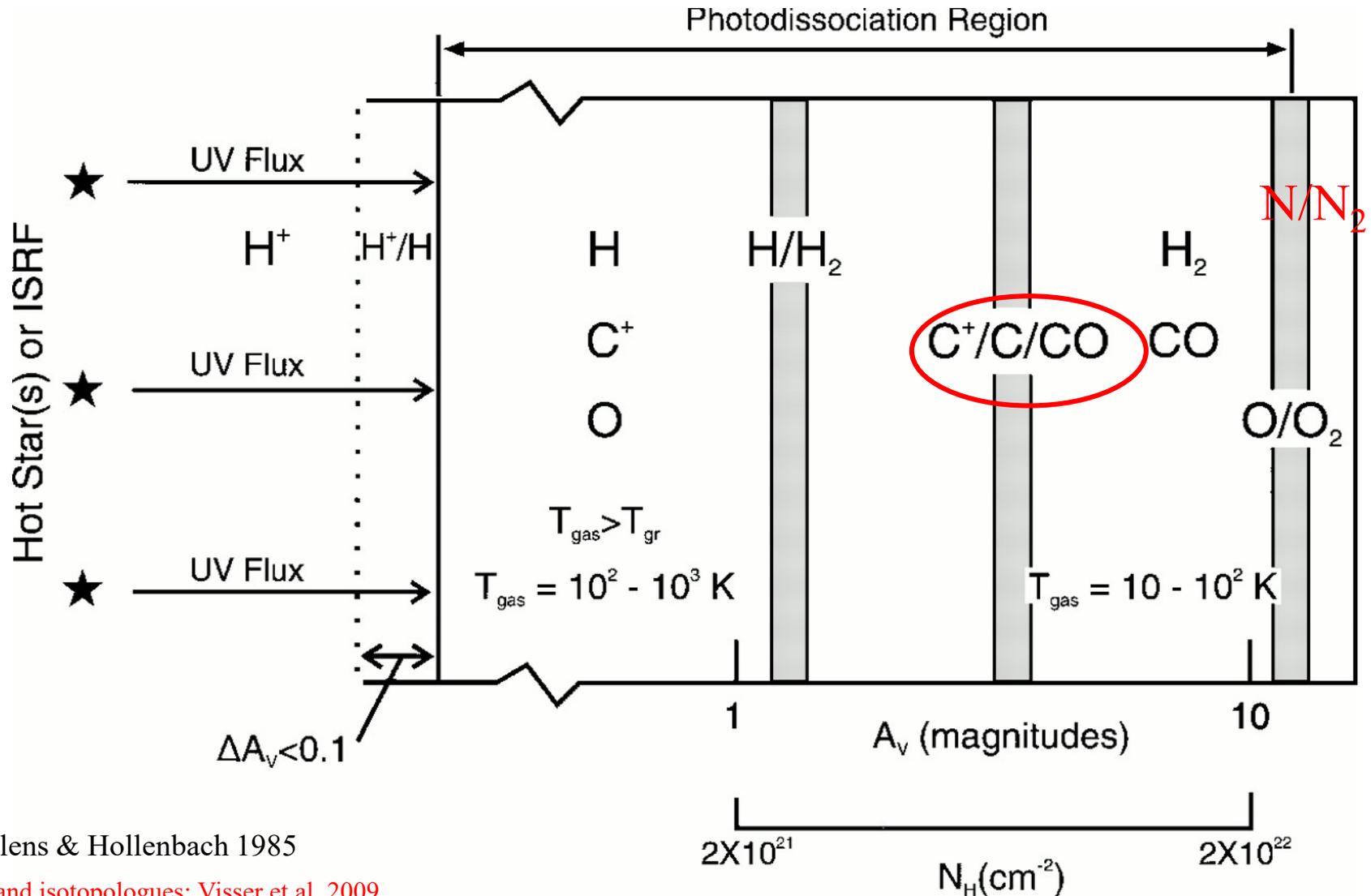
ESO PR Photo 02a/02 (25 January 2002)

© European Southern Observatory



AAT

PDR structure

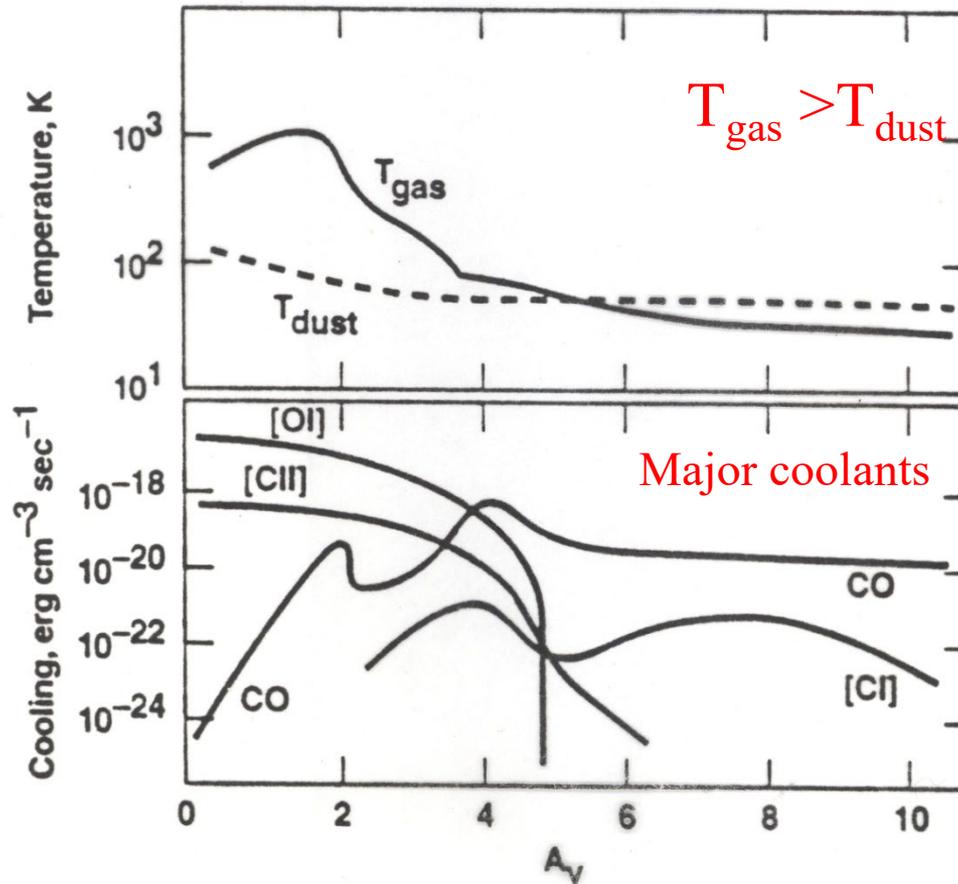


Tielens & Hollenbach 1985

CO and isotopologues: Visser et al. 2009

Hollenbach & Tielens 1997

Orion PDR physical structure

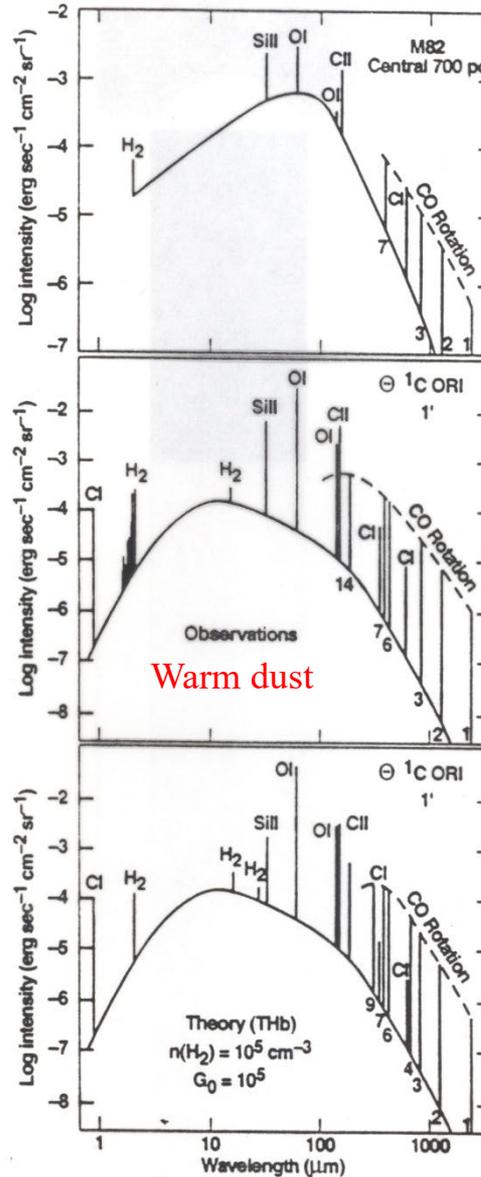


$n_{\text{H}}=10^5 \text{ cm}^{-3}$
 $I_{\text{UV}}=10^5$

Tielens & Hollenbach 1985

High temperature at edge allows $\text{O} + \text{H}_2 \rightarrow \text{OH} + \text{H}$ and $\text{C}^+ + \text{H}_2 \rightarrow \text{CH}^+ + \text{H}$ reactions to proceed, in contrast with low density diffuse clouds

Global PDR emission

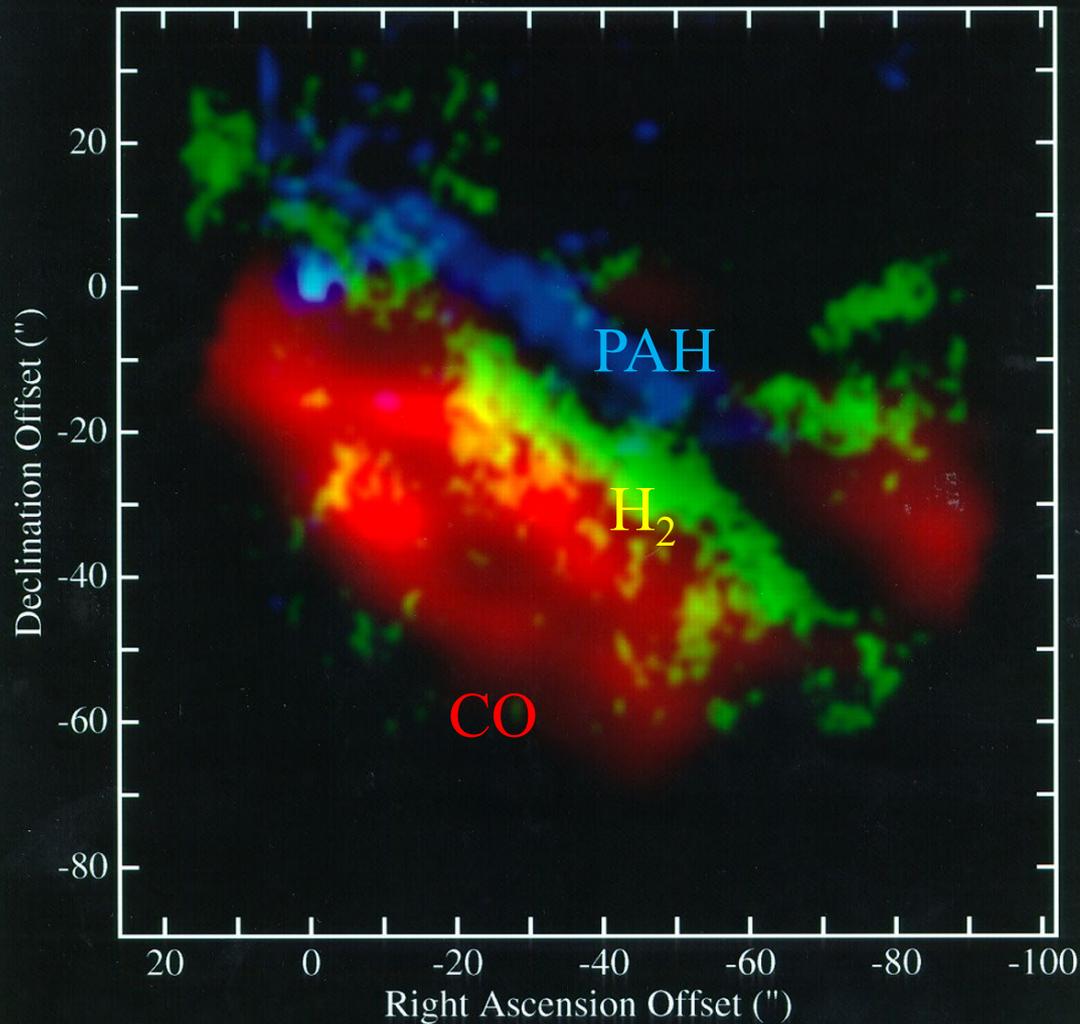


M 82 galaxy

Orion

Dense PDR model

Orion Bar PDR



Yellow: H₂ v=1-0

Blue PAH

Red: CO

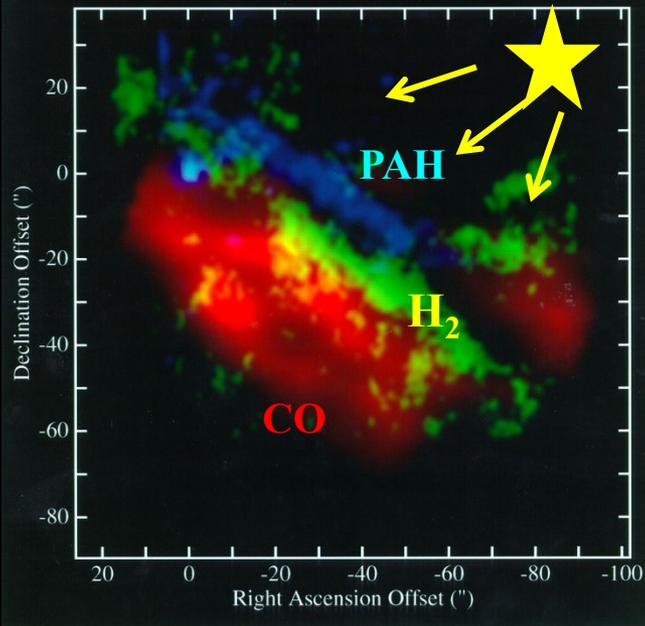
(0,0)= Θ^2 A Ori

Note layered structure!

Tielens et al.
1993

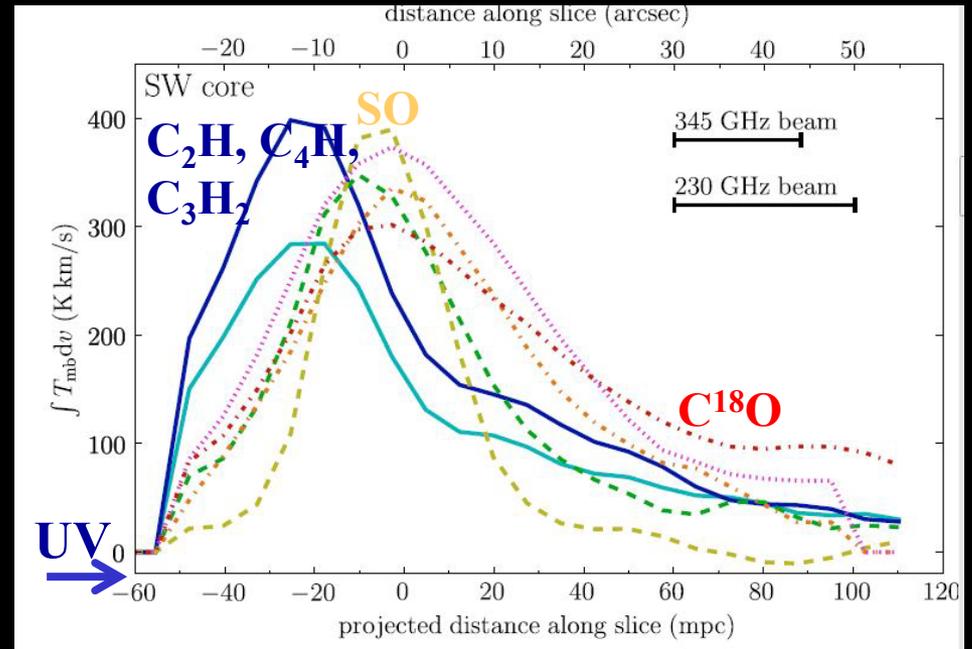
Layered structure PDRs

Orion Bar



Tielens et al. 1993

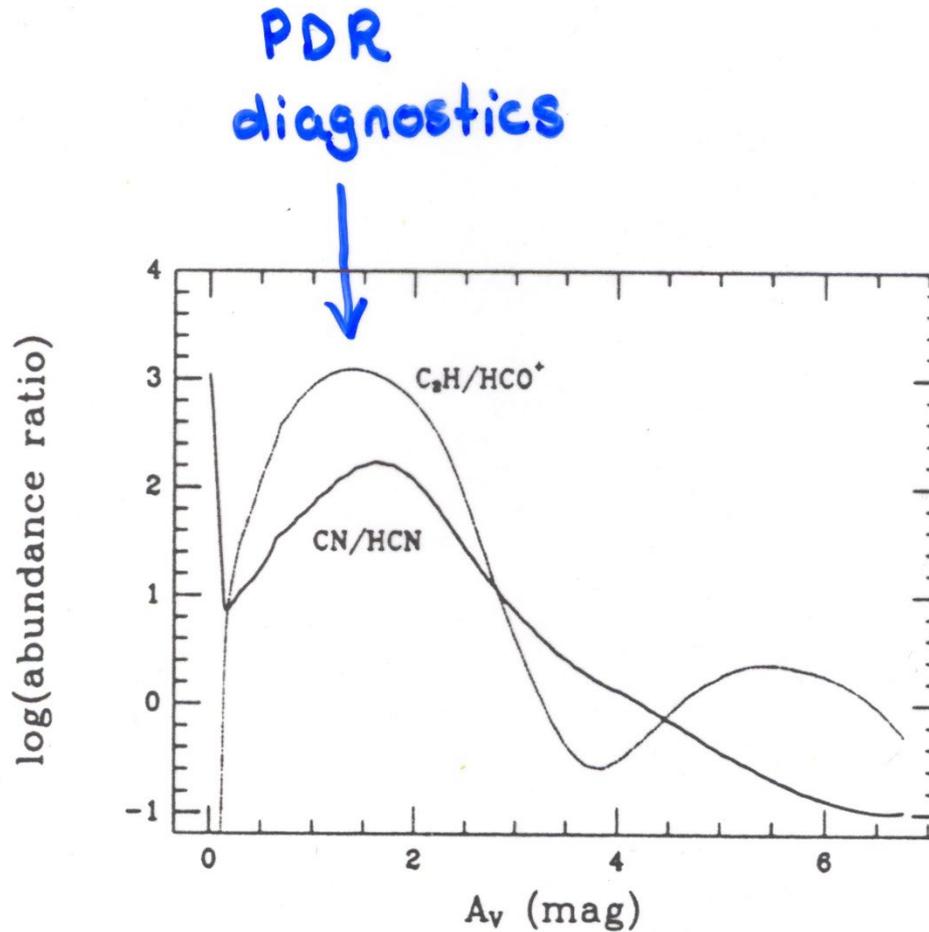
Radicals peak ahead



Hogerheijde et al. 1995, van der Wiel et al. 2009

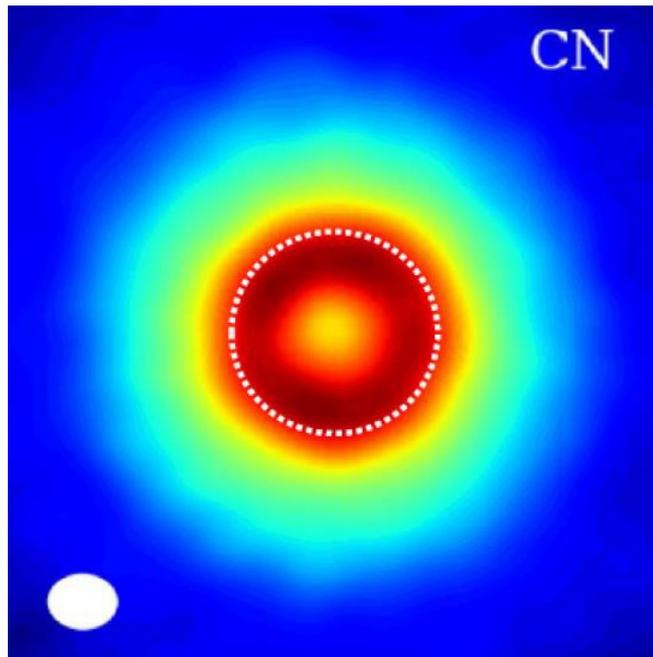
- Radicals C₂H, CN peak at C⁺ → C transition
- Stable HCN, CO peak deeper into cloud

PDR diagnostics: CN/HCN



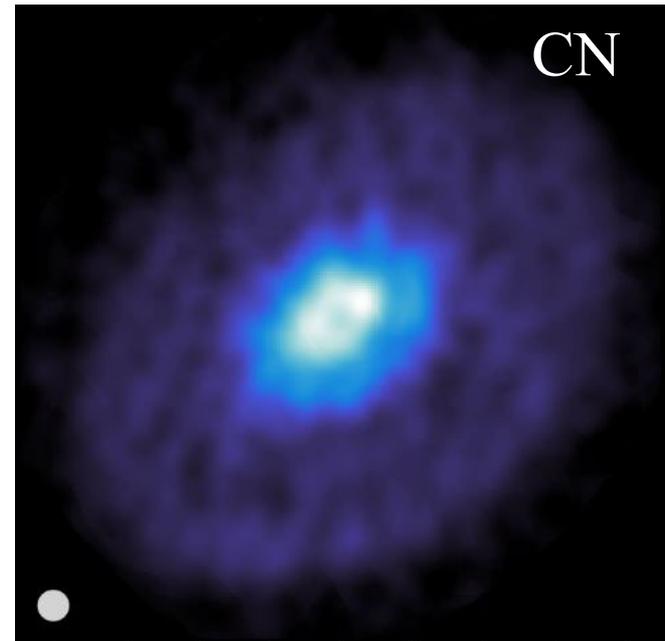
Jansen et al. 1995

4.5 Other examples PDRs: Protoplanetary disks



TW Hya disk

Teague et al. 2016,
Cazzoletti et al. 2018



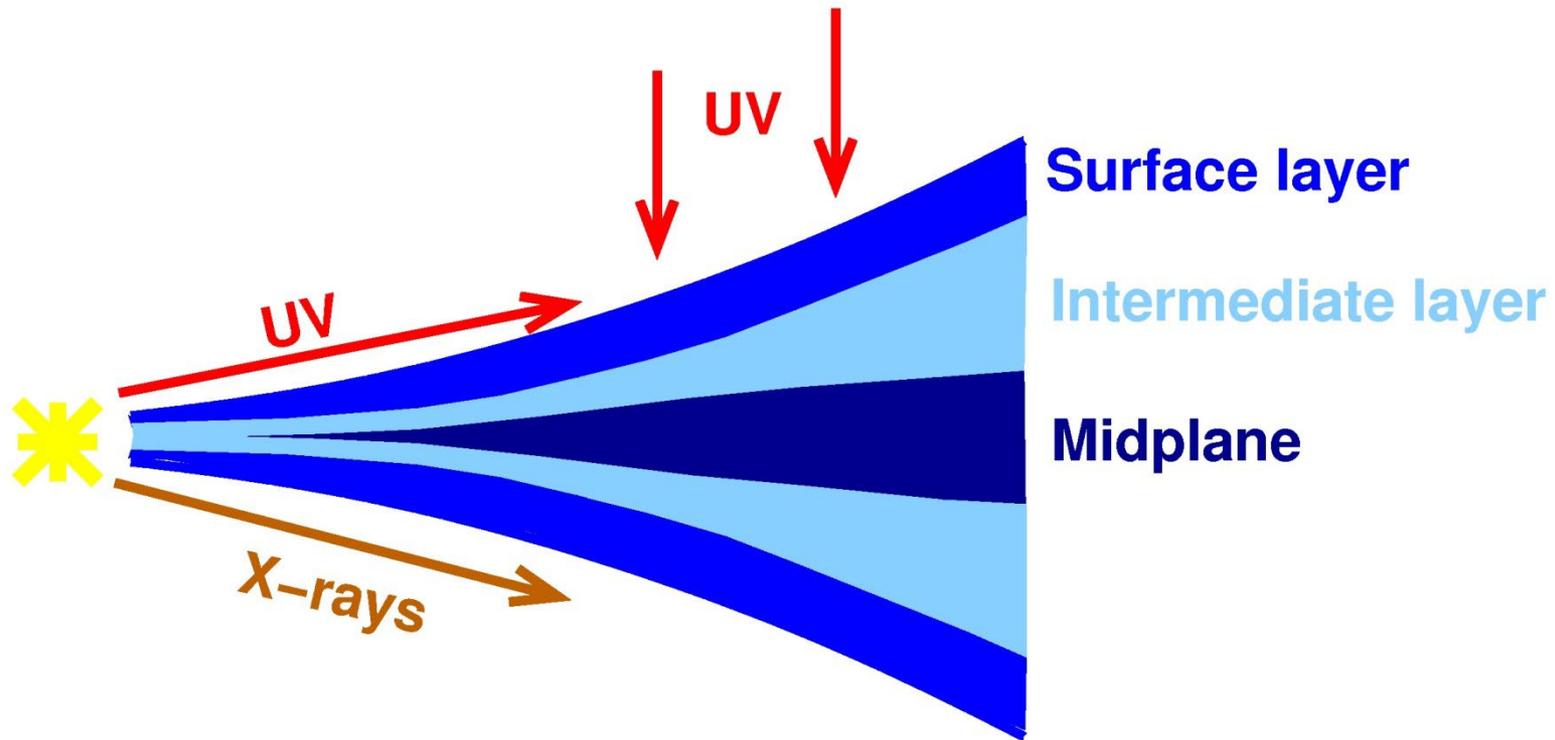
HD 163296 disk

Bergner et al. 2021, MAPS

Ring-like CN emission on ~ 100 AU scales

2D flaring protoplanetary disks

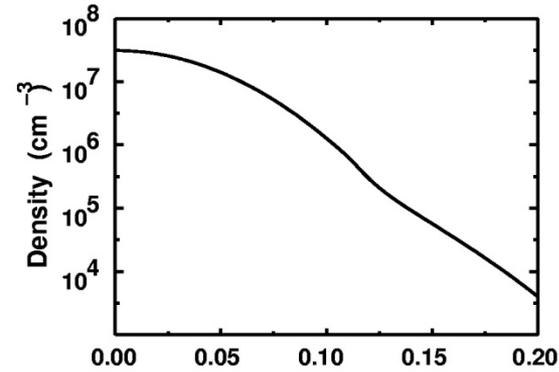
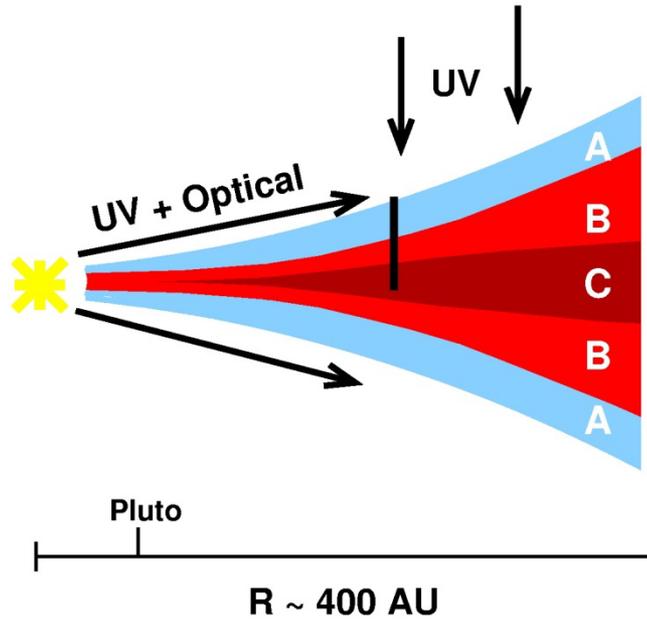
- Surface layer: molecules dissociated by UV photons
- Warm intermediate layer: molecules not much depleted, rich chemistry
- Cold midplane: molecules heavily frozen out



PDR-like structure but with large density gradients
and very intense UV (up to 10^6 x ISM)

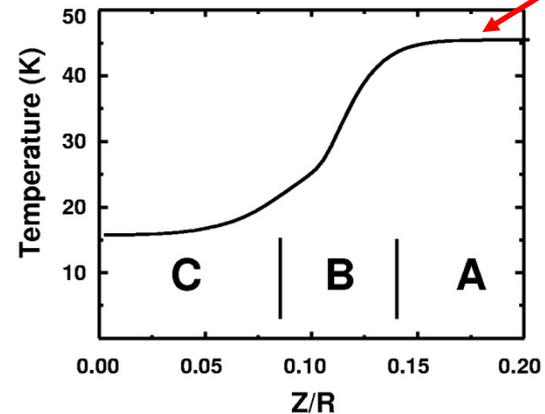
Aikawa et al. 2002
Bergin et al. 2007, PPV
+ many other groups

Vertical structure



R=200 AU

T_{gas} is larger

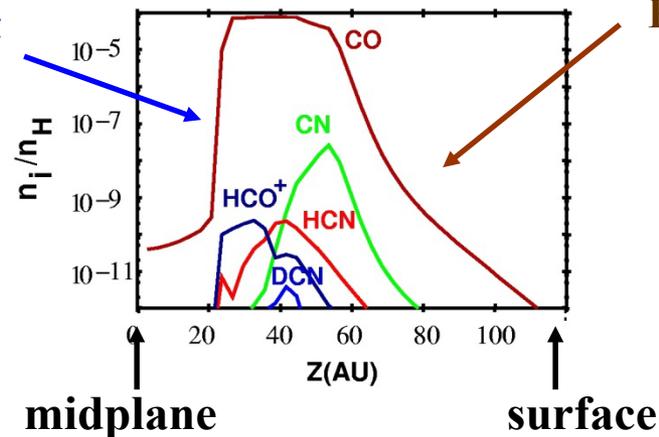


Freeze-out

Photodissociation

Ionization fraction:

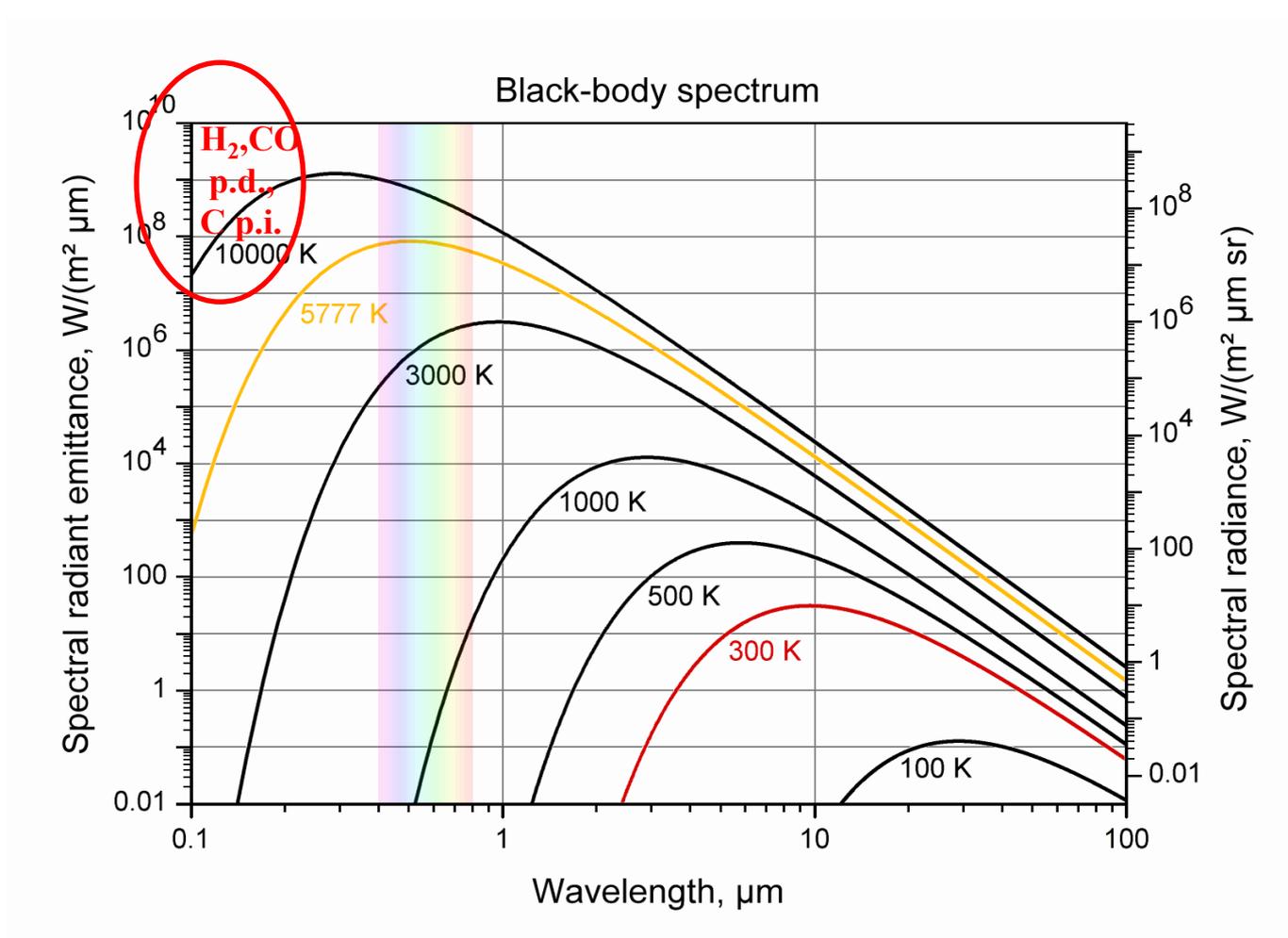
- Surface: 10^{-4} (C^+)
- Intermediate: 10^{-9} (HCO^+)
- Midplane: $\sim 10^{-11}$ (H_3^+ , H_2D^+)



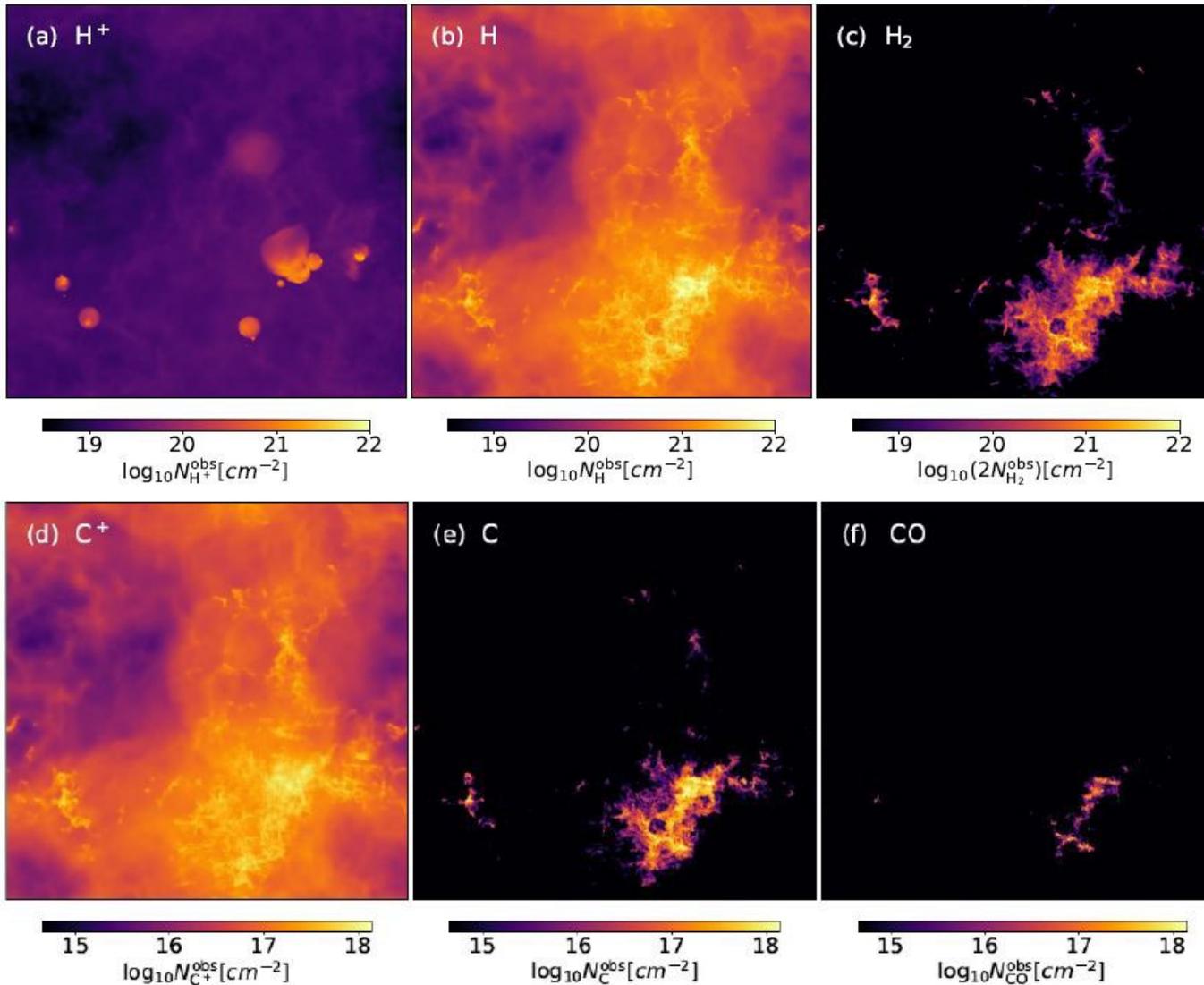
Beyond the standard model

- Cool PDRs
 - Clouds exposed to A, F, ... stars with $T_{\text{eff}}=6000-10000$ K \Rightarrow fewer photons between 11 and 13.6 eV to dissociate H_2 and CO
- Metal poor PDRs
 - Regions with C, O, ... depleted by factors >4 . Large gas/dust ratios (ex: LMC, SMC)
- Inhomogeneous, clumpy PDRs
 - Regions with large density contrasts and small filling factors clumps
- Time-dependent PDRs
 - Regions with evolving radiation field (e.g., planetary nebulae) or flowing gas (e.g., expanding HII region)
- XDRs
 - Regions exposed to X-rays, which can ionize H, He and multiply ionized atoms (e.g., S^{++}). Penetrate farther into cloud than UV

Cool radiation fields



Extragalactic multiphase ISM



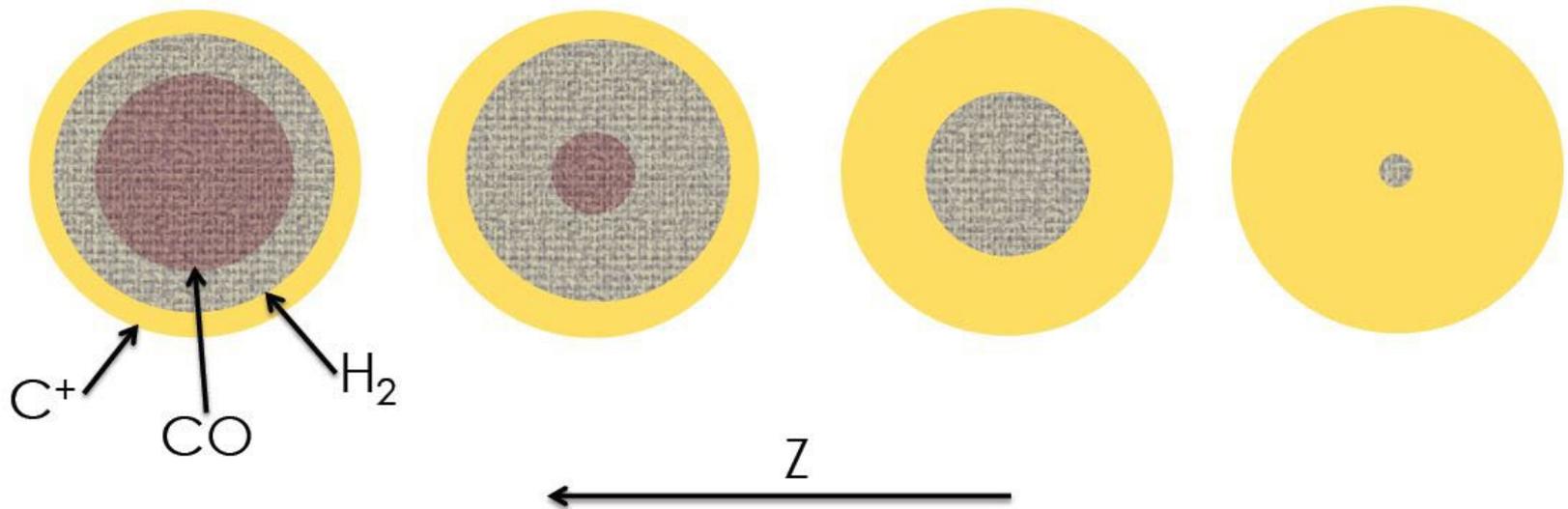
kpc scales

Incl. supernova
feedback

**Note CO limited
to small patch**

Hu, Sternberg, vD 2021

Low metallicity ISM



Pak et al. 1998
Wolfire et al. 2010

- At low metallicity, less dust and less C \rightarrow less shielding
- Larger fraction of the cloud is atomic at low Z

Summary Lecture 4

- Diffuse and translucent clouds are unique laboratories for testing gas-phase chemistry
- New data and insight from mm+FIR absorptions
 - Herschel, ALMA
- Several successes, but also some problems
 - Sometimes physics rather than chemistry not understood (e.g., turbulence)!
- Structure diffuse and translucent clouds as well as dense PDRs controlled by $\text{H} \rightarrow \text{H}_2$, and $\text{C}^+ \rightarrow \text{C} \rightarrow \text{CO}$ transitions

Appendix: equivalent width of spectral lines

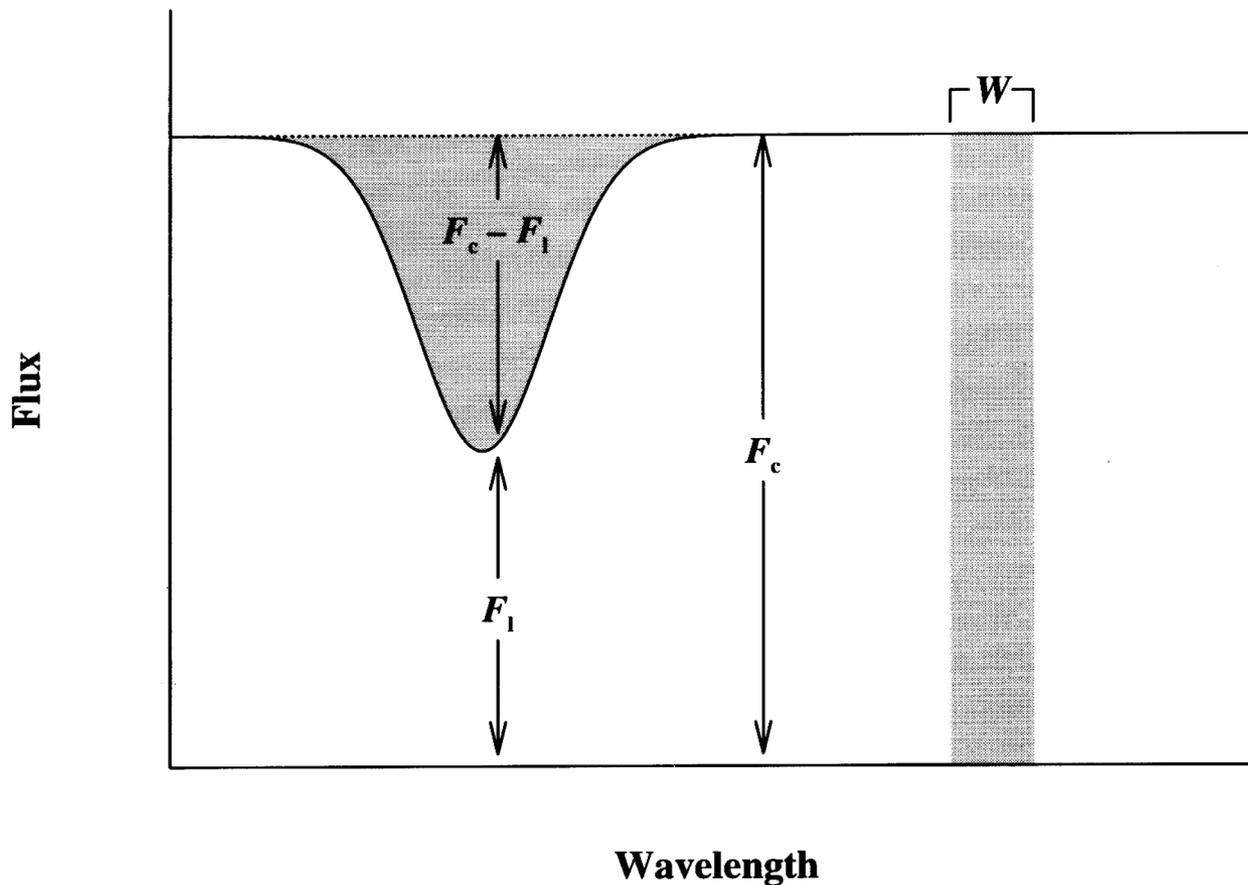
- In practice, resolution at optical wavelengths often insufficient to resolve line \Rightarrow measure only line strength or equivalent width
- Definition of equivalent width of line:

$$W_\nu = \int_{-\infty}^{\infty} (1 - e^{-\tau_\nu}) d\nu = \int_{-\infty}^{\infty} \left[\frac{I_\nu(0) - I_\nu}{I_\nu(0)} \right] d\nu \quad \text{Hz}$$

- W_ν is the width of a rectangular profile from 0 to $I_\nu(0)$ that has the same area as actual line
- W_ν measures line strength, but units are Hz
- In wavelength units

$$W_\lambda = \frac{d\lambda}{d\nu} W_\nu = \frac{\lambda^2}{c} W_\nu \quad \text{cm}$$

Schematic drawing of equivalent width of line



Column density $N \propto W/f$ if line optically thin
with f = oscillator strength